

# Biotechnological valorization of agro industrial and household wastes for lactic acid production

## Valorización biotecnológica de residuos agroindustriales y domésticos para la producción de ácido láctico

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### ABSTRACT

Lactic acid (LA) is an organic compound used in several industries, such as food, textile, chemical, and pharmaceutical. The global interest in this product is due to its use for the synthesis of numerous chemical compounds, including polylactic acid, a biodegradable thermoplastic and substitute for petroleum-derived plastics. An in-depth overview of the use of industrial and household wastes as inexpensive substrates in order to reduce the cost of LA production is presented. A review is carried out of the biotechnological aspects that must be taken into account when using some wastes with high transformation potential to produce LA in a submerged culture, as well recommendations for their use. The advantages and disadvantages of different types of treatments used for the transformation of waste into suitable substrates are considered. Several methods of fermentation, as well as genetic strategies for increasing the production, are summarized and compared. It is expected that in a few years there will be many advances in these areas that will allow greater large-scale production of LA using agroindustrial or household wastes, with potential positive economic and environmental impact in some regions of the planet.

**Keywords:** Lactic acid, microorganism, cellulose, starch, hydrolysis, organic waste.

### RESUMEN

El ácido láctico (AL) es un compuesto orgánico utilizado en diferentes industrias como la alimentaria, textil, química y farmacéutica. El interés mundial en este producto se debe a su uso para la síntesis de numerosos compuestos químicos, entre los que se incluye el ácido poliláctico, un termoplástico biodegradable y sustituto del plástico derivado del petróleo. En este artículo se presenta una descripción general y en profundidad, del uso de residuos agroindustriales y domésticos como sustratos económicos para reducir los costos de producción del AL. La revisión aborda los aspectos biotecnológicos que deben ser considerados al utilizar algunos residuos con alto potencial de transformación para producir AL en un cultivo sumergido, así como algunas recomendaciones para su uso. Además, se consideran las ventajas y desventajas de diferentes tipos de tratamientos empleados para la transformación de residuos en sustratos adecuados. Finalmente, se resumen y comparan varios métodos de fermentación, así como estrategias genéticas para incrementar la producción de ácido láctico. Se espera que en pocos años existan más avances en esta área, que permitan una mayor producción de AL a gran escala usando residuos agroindustriales y domésticos, con un impacto económico y ambiental positivo, en algunas regiones del planeta.

**Palabras clave:** ácido láctico, microorganismos, celulosa, almidón, hidrólisis, residuos orgánicos.

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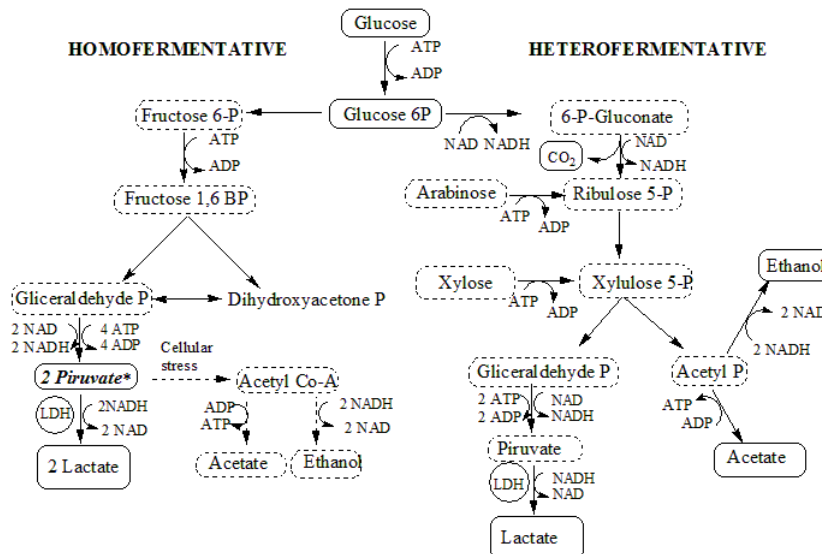
## INTRODUCTION

Lactic acid (LA) is an important naturally-occurring hydroxycarboxylic acid used in industry today (Castillo Martinez *et al.*, 2013). It is valuable because of its versatile applications, mainly in food as a preservative and as an acidulating agent, and to a lesser degree in the pharmaceutical, cosmetic, and chemical industries. Due to its carboxylic and hydroxyl groups, LA is highly reactive and participates in numerous chemical reactions, yielding important products such as pyruvic acid, acrylic acid, lactide, and lactate ester (C. Gao, Ma, & Xu, 2011). It exhibits optical activity with L (+) and D (-) isomers. However, it is mainly found in the inactive L and D form made up of equimolar fractions of each isomer (Boonpan, Pivsa-art, Pongswat, Areesirisuk, & Sirisangsawang, 2013). Given that humans have only the L(+) lactate dehydrogenase enzyme, the L(+) LA isomer is preferred for commercial applications involving human consumption (Bishai, De, Adhikari, & Banerjee, 2013). Some years ago, the potential of LA as a precursor of polylactic acid, a biopolymer that is biodegradable and biocompatible, was discovered. This polymer has been used in medical applications, 3D printing, and the fabrication of plastics (RedCorn, Fatemi, & Engelberth, 2018). Because of this, the LA market has grown considerably, and the global demand is expected to increase by around 14% to 19% per year up to 2020, generating expectations for the improvement of current production processes in order to meet this demand (Hu, Kwan, Daoud, Sze, & Lin, 2017). Presently, the global production of lactic acid is achieved mainly through fermentation (Abdel-Rahman, Tashiro, & Sonomoto, 2011). Nevertheless, a big difficulty for the fermentation process is

the high cost of the raw materials used as sources of energy and carbon, mainly pure sugars, representing an operational cost between 30% and 60% of total production costs (Marques, Santos, Francisco, & Roseiro, 2008; Oh *et al.*, 2005). Biotechnological processes can help to reduce the production costs by using organic waste from agricultural and food industries or from households (Bilanovic, Chang, Isobaev, & Welle, 2011). In addition to increasing the economic feasibility of the process, the use of waste in the fermentation is also an alternative that allows minimizing the impact caused by the dumping of waste into the environment (RedCorn & Engelberth, 2016). This review gives an overview of industrial and household wastes for utilization in the biotechnological production of lactic acid, aspects of fermentation for each waste, mainly in submerged culture, the advantages and disadvantages of different types of treatments used for the transformation of waste into suitable substrates, and some recent strategies for improving yields. In addition, specific recommendations are given for enhancing the value of these raw materials.

## FERMENTATIVE PATHWAY AND LA-PRODUCING MICROORGANISMS

LA can be produced by fermentation in one or two stages, according to the feedstock and type of microorganism used. The use of refined sugars, the addition of enzymes, co-culture systems, and microorganisms with the enzymatic ability to degrade complex substrates may allow carrying out the fermentation in a single step. The process with agroindustrial and household wastes can also occur in two steps. First, a pretreatment must be carried out in order to release sugars into the medium, which is generally enzymatic or thermochemical, and



**Figure 1.** Metabolism of lactic acid bacteria - homofermentative and heterofermentative pathway. (Adapted from Hofvendahl & Hagerdal, 2000).

second, these sugars are used in the bioreactor for the fermentation and production of LA (John, Anisha, Nampoothiri, & Pandey, 2009). Lactic acid bacteria (LAB) and filamentous fungi have proven to be the most attractive microorganisms for LA production because of their high productivity and high yield (Nuttha, 2005).

### Lactic acid bacteria

LAB is a large group of benign microorganisms that produce LA during anaerobic fermentation. This group can be divided into homofermentative and heterofermentative bacteria, according to the nature of the final products, as is shown in figure 1. The homofermentative bacteria metabolize hexoses via the Embden-Meyerhof pathway (glycolysis), generating two molecules of LA for each molecule of glucose, but they do not ferment pentoses or gluconate.

The heterofermentative lactic acid bacteria metabolize hexoses, pentoses such as xylose, and some other substances, following the phosphoketolase pathway (Panesar, Kennedy, Gandhi, & Bunko, 2007). The difference with the homofermentative route is the conversion of two of the six carbons of glucose to acetic acid, which limits yields and increases the costs of LA purification (Cock & Rodríguez, 2005). The proportion of the by-products is dependent on the redox potential of the system. In figure 1, it is shown that homofermentators can also generate mixed acids, but this only occurs when the microorganism is subjected to adverse culture conditions (Hofvendahl & Hagerdal, 2000).

### Fungi

Filamentous fungi of the genus *Rhizopus* have been recognized for producing L(+) LA efficiently via aerobic metabolism and with lower nutritional requirements than LAB. These latter organisms need vitamin B complex, several aminoacids, and nucleotides for their growth (Abdel-Rahman, Tashiro, & Sonomoto, 2013; Watanabe *et al.*, 2012). This fungus can produce other organic acids besides L(+) LA, such as fumaric acid, malic acid, and ethanol. Glucose is degraded to pyruvate through the Embden-Mayerhof-Parnas pathway, and it is used as a precursor for the different metabolites (Nuttha, 2005). The conversion of xylose to lactic acid has not been investigated as thoroughly as the conversion of glucose. However, it has been found that *Rhizopus oryzae* is also capable of metabolizing xylose via pentose phosphate to produce xylulose 5 phosphate, which subsequently can be metabolized via the Embden-Meyerhoff-Parnas pathway (Z. Y. Zhang, Jin, & Kelly, 2007).

These fungi can degrade more complex substrates than LAB, and their heterogeneous morphology enables easy

separation of the biomass (Saito, Hasa, & Abe, 2012). Fungi in a submerged culture may take different physical forms: small pellet morphology is reported to be appropriate for LA fermentation, while the mycelial form causes problems of poor mass transfer and low production of LA (Liu, Liao, Liu, & Chen, 2006). Tools such as cell immobilization have been implemented in order to control the morphology and increase the production of LA (Wang, Wang, Yang, Wang, & Ren, 2010). In spite of the nutritional and separation advantages of LA in a bioprocess with fungi, we consider that the use of LAB on an industrial scale is more feasible, due to the easier operation of bioreactors. In addition, LAB have been more thoroughly studied, so they have relative advantages for increasing LA production through genetic transformations.

### PRETREATMENT METHODS

One of the biggest problems for the use of agroindustrial and household wastes for the production of lactic acid at the industrial level is the difficulty of achieving an integral and efficient use of all the nutrients of these raw materials, because they have a complex structure. Therefore, it is necessary to perform a pre-pretreatment of the raw material before the fermentation. The main objective of the pretreatment of substrates such as starch or lignin is the gradual breakdown of the recalcitrant polysaccharides and matrices into low-weight molecules easily digestible by microorganisms. Currently, various pretreatments have been used, among them physical ones, such as milling and grinding; chemical ones, such as acid and alkaline hydrolysis; physical-chemical ones, such as steam explosion and hydrothermolysis; and biological ones, such as the use of fungi, bacteria, or enzymes (Maria & Valencia, 2011). Complex materials usually need a physical treatment in order to reduce the size of the particles, so that the residue will be manageable and have better exposure to subsequent pretreatments. Acid and enzymatic hydrolysis are the most-reported treatments in studies of residue utilization for application in fermentative processes. Due to this, two types of pretreatment will be analyzed.

#### Acid hydrolysis

Acid hydrolysis is based on the use of acids (sulfuric, hydrochloric, trifluoroacetic, or hydrosulfuric) to perform catalysis at high temperatures and break complex matrices (Abdel-Rahman *et al.*, 2011). The effectiveness of the waste degradation depends on three interdependent variables: acid concentration, temperature, and exposure time (Domínguez *et al.*, 2011). Acidic conditions between 0.5% and 1.0% are considered low, and those above 1.5% strong; temperatures near 121 °C and 160 °C are used. Nevertheless, in order to decrease the degradation of sugars such as fructose and sucrose, which

occurs at 106 °C in acidic environments, temperatures below 120 °C are employed in the industry (Maria & Valencia, 2011). Hydrolyzed product may contain sugars such as glucose, xylose, and arabinose, but their proportion in the medium depends on the type of substrate used and the pretreatment conditions (M. Gao, Kaneko, Hirata, Toorisaka, & Hano, 2008; Laopaiboon, Thani, Leelavatcharamas, & Laopaiboon, 2010). In most cases, the different types of sugars released are not all consumed by a single organism, decreasing the product yield. Besides producing sugars in this process, some soluble materials such as lignin, acetic acid, furfural, 5-hydroxymethylfurfural (HMF), levulinic acid, and phenolic compounds can also be generated. These compounds are toxic and may inhibit both the growth and the sugar utilization capabilities of the microorganisms (Laopaiboon *et al.*, 2010). The phenomenon of inhibition in LA fermentation from acid hydrolyzates has been reported to occur with microorganisms such as *L. casei* subsp. *rhamnosus*, *L. pentosus*, and *E. coli* (Guo, Jia, Li, & Chen, 2010). In order to reduce the inhibitory effect, the hydrolyzate is subjected to detoxification through methods that include, among others, neutralization with alkali or activated carbon and ion exchange (Woiciechowski, Soccol, Ramos, & Pandey, 1999). Woiciechowski *et al.* (1999), found that *R. oryzae* NRRL 395 has the ability to resist the inhibitory effect of toxic components in the hydrolyzate. Guo *et al.* (2010), isolated a microorganism identified as *L. brevis* (S3F4), which is able to tolerate potential inhibitors present in the hydrolyzate of corn stover. The use of acid hydrolysis with complex substrates also can destroy proteins and vitamins, which are essential for the growth and LA production of LAB.

### Enzymatic hydrolysis

Enzymatic hydrolysis is the most promising method for obtaining a good yield of fermentable sugars from organic waste. Enzymes such as laccase, lignin peroxidase,

and manganese peroxidase are used to degrade lignin; endoxylanases, acetyl xylan esterase,  $\beta$ -xylosidase,  $\beta$ -mannosidase, and  $\alpha$ -L-arabinofuranosidase are utilized in the saccharification of hemicellulose; while cellobiohydrolases,  $\beta$ -glucosidases, and endoglucanases are employed for cellulose degradation. The use of individual enzymes involves limitations, while enzyme mixtures can exhibit a chain reaction that increases the effectiveness of the hydrolysis (Van Dyk & Pletschke, 2012). On the other hand, for hydrolyzed starch, the enzymes  $\alpha$ -amylase,  $\beta$ -amylase, and glucoamylase are used. These kinds of enzymes attack  $\alpha$ -1.4 links but not  $\alpha$ -1.6 links, which are hydrolyzed by isoamylases or dextrinases. The enzymatic process for degrading starch generally takes place in two consecutive stages: liquefaction and saccharification. Amylase acts during the first step, and glucoamylase acts during the second one. These processes are usually performed separately, because temperature conditions are different for each enzyme (Castaño & Mejia, 2008). In table 1, a comparison of the advantages and disadvantages of the use of acid and enzymatic pretreatment is shown. These two methods, acid and enzymatic hydrolysis, are equally effective when parameters such as percent of hydrolysis, time, cost of chemicals, and energy consumption are considered in one or another method. For example, in the hydrolysis process of 150 kg of cassava bagasse, US \$34.27 was required for acid hydrolysis, while the cost of enzymatic hydrolysis was US \$2470.99. Although acid pretreatment is less expensive and faster, it does not generate yields that are as high as does enzymatic hydrolysis (Woiciechowski, Nitsche, Pandey, & Ricardo, 2002). Therefore, the pretreatment selection depends on the substrate composition and the availability of resources. Presently, the American company NatureWorks LLC is the world leader in the production of lactic acid via biotechnology. The treatment of lignocellulosic raw materials by this company is done via the enzymatic method (NatureworksLLC, 2019).

**Table 1.** Comparison between acid and enzymatic hydrolysis.

	Advantage	Disadvantage
ACID HYDROLYSIS	<ul style="list-style-type: none"> <li>- Low cost of chemical</li> <li>- Fast processing</li> <li>- Ease of operation</li> </ul>	<ul style="list-style-type: none"> <li>- Produces toxic compounds</li> <li>- Degradation of sugars</li> <li>- It adjustment of the pH is necessary before fermentation</li> <li>- The remaining acid residues can present a serious environmental problem</li> </ul>
ENZIMATIC HIDROLYSIS	<ul style="list-style-type: none"> <li>- Obtaining high concentrations of sugars due to the specificity of hydrolytic enzymes</li> <li>- No toxic compounds</li> <li>-Environmentally friendly process</li> </ul>	<ul style="list-style-type: none"> <li>- High cost of enzymes</li> <li>- Slow process</li> </ul>

## INEXPENSIVE SUBSTRATES FOR LACTIC ACID PRODUCTION

The highest concentration of LA is obtained when pure sugars are fermented. Glucose is considered to be the main raw material used in LA fermentations, because this sugar gives high yields and its assimilation is easy for the microorganisms. Nevertheless, the use of expensive substrates is a problem, since LA is a low-priced product (Oh *et al.*, 2005). The utilization of cheap raw materials such as organic waste has emerged as an alternative that overcomes this problem. Globally, solid wastes have a negative environmental impact, caused by their improper disposal and the increasing rate at which they are generated, associated with industrial growth and human abuse of the environment (María & Echeverri, 2004; Pleissner *et al.*, 2017). Alarm over high rates of pollution has led to a search for new processes that would take advantage of organic waste before its being discarded. Agroindustrial and household wastes have shown great potential for biotechnological processes on a large scale, because they are abundant, are low-priced, and have a high nutrient content that can be transformed into a valuable product (Yousuf, Bastidas-Oyanedel, & Schmidt, 2018). Optimal raw materials for this purpose should have low levels of pollutants and toxic substances, be easily fermentable, need little or no treatment, and generate high levels of production of the metabolite of interest (Y.-J. Wee & Ryu, 2009). The residues used for LA production can be divided into three groups, according to their composition: lignocellulosic waste, starchy waste, and residues from dairy processes.

The production policy of NatureWorks LLC focused on the use of renewable and abundant raw materials such as corn, cassava, sugarcane, and beets. However, in

2016 they inaugurated a new laboratory in order to implement a fermentation technology that allows the transformation of methane (greenhouse gas) to lactic acid. Other companies, such as CORBION (formerly called PURAC), Galactic S.A, Hefei TNJ Chemical, and Shanghai Honghao Chemicals Co. are also recognized producers of lactic acid through fermentation processes (Corbion, 2019; Galactic, 2019; NatureworksLLC, 2019).

### Lignocellulosic material

Lignocellulose is derived from organic materials and represents the most abundant global source of biomass that has not yet been effectively exploited. It is predominantly composed of a thoroughly mixed matrix of cellulose, hemicellulose, and lignin, as well as lesser amounts of minerals, oils, and other components (Doherty, Mousavioun, & Fellows, 2011). Each lignocellulosic material has a unique chemical composition. The composition in the matrix is variable, in accordance with species, age, growth stage, and environmental conditions (Cuervo, Folch, & Quiroz, 2001), as is shown in table 2.

Cellulose is the major component of plant biomass, which is constantly regenerated by photosynthesis. It is a rigid homopolysaccharide formed by linear units of  $\beta$ -D glucans. Hemicellulose is a heteropolysaccharide that is composed of two pentose sugars (xylose and arabinose) and three hexose sugars (glucose, mannose, and galactose). This polymer is more easily degraded by microorganisms than cellulose and lignin (Guo *et al.*, 2010). Finally, lignin is a functional material that gives structural strength and rigidity to plants. Physically, it is a macromolecular aromatic polymer formed by the union of phenylpropane monomers that generates an extremely com-

Table 2. Different organic wastes with content of cellulose, hemicellulose and lignin.

Biomass	Cellulose (%)	Hemicellulose (%)	Lignin (%)	References
Bagasse	38.10 - 43.00	25.00 - 27.20	12.20 - 24.00	(Merino & Cherry, 2007; Van Dyk & Pletschke, 2012)
Corn cobs	35 - 45.00	35 - 42	4.5 - 15.00	(Howard, Abotsi, L, & Howard, 2003)
Hardwood	40-55	24-40	18-25	(Sun & Cheng, 2002)
Leaves	15-20	80-85	0.00	(Howard <i>et al.</i> , 2003)
Wheat Straw	30.00 - 44.00	23.00 - 50.00	10.40 - 17.00	(Abdel-Rahman <i>et al.</i> , 2011; Merino & Cherry, 2007)

plex structure. Thus plants that have high amounts of lignin in their composition are more resistant to attacks by microorganisms and to some extreme conditions (Abdel-Rahman *et al.*, 2011). Due to low bioavailability of sugars in lignocellulosic biomass, these materials are subjected to a pretreatment in order to liberate fermentable sugars before fermentation (Maas, Bakker, Eggink, & Weusthuis, 2006). Next, some work done on the production of LA with lignocellulose residues will be described.

**Sugar cane.** Since 2005, in Colombia five of the thirteen sugar manufacturers that exist in the country have distilleries dedicated to the production of fuel alcohol through biotechnological processes. In 2011, 337 million liters of bioethanol were produced from sugarcane. The growing use of this crop for the biofuel industry has diverted a high percentage of the raw material previously used for sugar production, entering into direct competition with the generation of this food, which is in high demand nationally. Characteristic residues of this gramineae have been investigated for use in bioprocess applications (Asocaña, 2012). Bagasse, which is generated after trituration and extraction of juice from sugar cane, is considered to be the main by-product from sugar production centers (Maria & Valencia, 2011). This inert bagasse has been investigated along with cassava hydrolyzate for carrying out LA fermentation in the solid state (Rojan, Nampoothiri, Nair, & Pandey, 2005). In submerged cultures, Laopaiboon *et al.* (2010), obtained a LA concentration of 10 g/L, while Adsul *et al.* (2007), generated 67 g/L of the metabolite with a yield of 80% by using the genetically modified strain *L. delbrueckii* subsp *delbrueckii* mutant Uc-3. Cock and Rodriguez (2007) determined that a mixture of buds and leaf juice proved to be suitable as raw material for the growth of LAB, even without the addition of a nitrogen source. Molasses contains about 50% (w/w) of sugars; the most abundant of these is sucrose (Hofvendahl & Hagerdal, 2000). The high concentration of this sugar produces a viscous liquid that may bring on substrate inhibition. Also, molasses contains metals (iron, zinc, copper, etc.) that may affect the growth of the microorganisms and are implicated in the inactivation of the enzymes important for the biosynthesis of products (Y.-J. Wee, Kim, Yun, & Ryu, 2004). An advantage of this residue is its buffer capacity, which allows maintaining a pH above 5 during the fermentation of LA, avoiding the need to add neutralizers to the culture medium (Dumbrepatil, Adsul, Chaudhari, Khire, & Gokhale, 2008). Most of the results of LA production with molasses reported in the literature range between 40 and 166 g/L, with a volumetric productivity from 2.00 to 4.20 g/L h and a yield between 0.80 and 0.98 g/g (Cock & Rodríguez, 2005; C. Gao *et al.*, 2011; Pal, Sikder, Roy, & Giorno, 2009). *L. delbrueckii* is the most widely

used microorganism for this substrate, although there are reports of the use of *E. faecalis*, *R. oryzae*, and *E. coli* (Dumbrepatil *et al.*, 2008; Y. Wee, Kim, & Ryu, 2006).

Wheat bran and corn cobs. Wheat bran, in addition to being used as a carbon source, has also been shown to be an excellent source of nutrients containing nitrogen, with the possibility of replacing the expensive yeast extract used to provide nutrients for bacterial growth and fermentation (Li, Han, Ji, Wang, & Tan, 2010; Y. Wee *et al.*, 2006). However, some researchers suggest that the use of yeast extract in small amounts is necessary in order to avoid growth factor limitations if nutrients from waste materials are being used in fermentation with LAB. If no yeast extract is used, fermentation time increases, causing decreased productivity (K. Lee, 2005; Lu, Lu, He, & Yu, 2009). As a carbon source, it has been found that wheat bran is composed of lignocellulose and a considerable amount of starch; therefore, it is an interesting material for study. Investigations by Yun *et al.* (2004), show a LA production of 60.7 g/L from 300 g/L wheat bran. Garde *et al.* (2002), used wheat straw hydrolyzate and a co-culture of LAB to increase yields in the fermentation, while Bulut *et al.* (2004) reported that the use of wheat bran with *Rhizopus oryzae* is not sustainable if this substrate was used as the only carbon source. Similar to wheat bran, corn cobs are also an abundant agroindustrial waste rich in hemicellulose, cellulose, nitrogen, and protein. They can be used as a complete source of carbon and energy, or, as in the case of corn steep liquor, as a source of nitrogen (Oh *et al.*, 2005; Ruengruglikit & Hang, 2003). Miura *et al.*, (2004) have reported the use of *Rhizopus oryzae* with an *Acremonium* cellulase preparation (10 U/g corncorb). The authors obtained a concentration of total sugars of 55 g/L, composed mainly of glucose (61%) and xylose (21%) from the enzymatic treatment of 100 g/L of raw corncob. The yield for the conversion of glucose to lactic acid was 82%.

**Starchy material.** Starch is a polysaccharide that exists mainly in tubers, such as potatoes and cassavas, and in seeds and grains such as wheat, corn, and rice. This is a glucose polymer that contains long linear and branched chains of amylose and amylopectin, respectively (Woiciechowski *et al.*, 2002). This material must be hydrolyzed before fermentation, because its degradation is difficult, although it can also be fermented directly by LAB and fungi with amylolytic ability (Abdel-Rahman *et al.*, 2013; Watanabe *et al.*, 2012). LAB with amylolytic activity include *L. fermentum*, *L. plantarum*, *L. manihottivorans*, *L. amylovorus*, and *L. amylophilus*, among others (Reddy, Altaf, Naveena, Venkateshwar, & Kumar, 2008; Shibata, Flores, Kobayashi, & Sonomoto, 2007). In submerged fermentation with a different strain of *L. amy-*



*lophilus*, it has been reported that 90% of the soluble starch is efficiently converted to LA (Rojan *et al.*, 2005; Vishnu, Seenayya, & Reddy, 2002). In most studies of fermentation by *Rhizopus sp.* of starchy materials, they were converted directly, without pretreatment or the addition of supplements (Oda, Saito, Yamauchi, & Mori, 2002). The maximum conversion factor of starch to LA using *Rhizopus sp.* was 1.11 g of LA/g of starch (Huang, Jin, & Lant, 2005). Next, some work done on the production of LA with starchy material will be described.

**Cassava.** Cassava is one of the most abundant starch crops in the world, and it is the third largest source of food for the inhabitants of tropical regions (Teixeira *et al.*, 2012). This tuber grows in poor soils and contains more than 80% starch. Nonetheless, once collected, cassava deteriorates rapidly, which causes a great deal of reluctance to use it. Other disadvantages of this crop are its low protein content and its toxicity, due to the presence of cyanide in its peel (Teixeira *et al.*, 2012). Cassava bagasse has a high starch content, around 52% dry weight (Woiciechowski *et al.*, 2002). John *et al.* (2006), investigated the use of this residue with two LAB, *L. casei* NCIMB 3254 and *L. delbrueckii* NCIM 2025, reaching a maximum concentration of 83.80 g/L and 81.90 g/L of LA respectively. Later, John *et al.* (2007), employed a 1:1 mixture of the previously-mentioned strains to determine how their interactions affect the process, but only achieved a maximum product concentration of 81.00 g/L. With cassava powder (70% starch) as a carbon source, a yield of 0.85 g/g from 100.00 g/L of substrate was found. Similar results were obtained with glucose at 100.00 g/L and potato starch at the same concentration. Yin *et al.* (1997), found that 120.00 g/L is the optimum concentration of starch for fermentation by *R. oryzae*; high substrate concentrations exceeding 130 g/L result in low yields of LA.

**Potatoes.** Potatoes are one of the main starch crops in the world, with a production of  $2 \times 10^7$  t in the United States and  $9.5 \times 10^7$  t in China (RedCorn *et al.*, 2018). A quarter of this raw material is discarded during industrial processing, generating large volumes of waste, such as peels and wastewater, with high starch content (Z. Zhang, 2008). Potato peel has been shown to be an agroindustrial residue with great potential for use in the production of lactic acid, because it is a very complete substrate in relation to its composition of macro- and micronutrients. In addition, the process of bioconversion to LA is fast and efficient (RedCorn *et al.*, 2018). Bilanovic *et al.* (2011), made an economic assessment of the carbon sources used to produce LA. The price of potato waste is about 15 to 50 U.S. dollars/t, which is less than the cost of usual substrates in standard media

for LA production such as molasses, starch, lactose, and glucose (54 to 317 U.S. dollars/t). The use of this substrate could generate a recovery of 5600 million USD/year (RedCorn *et al.*, 2018). Ping *et al.* (2005), utilized potato wastewater and the fungi *R. oryzae* and *R. arrhizus* for LA production with direct fermentation. Both fungi demonstrated a high rate of starch hydrolysis. The temperature of 30 °C, pH 6 and initial starch concentration of 20 g/L were the best growing condition for obtaining a product/substrate yield of 0.87g/g with *R. oryzae* 2062 and 0.97 g/g starch with *R. arrhizus* 36017 after 36 h of fermentation. Potato starch was used by Oda *et al.* (2002), to assess 38 strains of *R. oryzae* in the production of this acid. *R. oryzae* 4707 produced the highest final concentration of LA and was reported to exhibit a gradual production of this metabolite during the first 6 days. Since this fungus was able to secrete amylases, cellulases, hemicellulases, and pectinases from 33.1% initial starch, only 7.6% starch remained in the culture medium at the end of fermentation. Zhang (2008) also used this residue and obtained 88.70 g/L LA from 100 g/L of substrate in a bubble column bioreactor with *R. arrhizus*.

**Rice.** Nearly 650 Mt of rice were produced worldwide in 2007. This amount of rice could also generate a large amount of rice bran, which represents between 8 and 10% of the weight of the grain (Li, Tianwei, Jike, Zixin, & Lu, 2012). Depending on the separation process, rice bran can be classified as yellow or white. They differ in the content of starch and oils. For LA fermentation, oil-free bran is preferred (Li *et al.*, 2012). Many researchers have worked on the feasibility of the production of LA from residues of rice production as a carbon source. However, some of them have also found considerable quantities of thiamine, riboflavin (B vitamins) and amino-acids in this raw material, and they have reported a positive result as a source of nitrogen and micronutrients (L. Gao *et al.*, 2008; M. Gao *et al.*, 2008). Watanabe *et al.* (2016), isolated a novel LAB (*L. rhamnusus* M23) able to produce extracellular protease with a high degree of activity. This enzyme can degrade complex substrates such as a non-sterilized mixture of the drainage from the washing of rice and rice bran, using them as the only source of nutrients for the growth and the production of LA. *L. delbrueckii* JCM1106, producer of D(-) LA, was used to transform the sugars present in the enzyme hydrolyzate from broken rice, achieving a high productivity of 3.59 g/Lh (Nakano, Ugwu, & Tokiwa, 2012). Lu *et al.* (2009), reported a yield of 73% in LA production using old untreated rice, which exceeds by 5.79% and 8.71% the production for fresh corn and processed rice, respectively. Most of the studies of LA production via fermentation consist of a single inoculation or a mixture of two microorganisms. Gao *et al.* (2008), focused on the utili-

zation of a LAB consortium consisting of *L. plantarum*, *L. fermentum*, and *L. paracasei* to increase the production of acid using rice straw, since *L. plantarum* and *L. paracasei* are homofermentative, while *L. fermentum* is heterofermentative. Not only LA, but also acetic acid, is obtained, at a ratio of 3.88:1 (lactic acid:acetic acid).

**Food Residues.** Food residues have a high content of moisture, carbon such as starch, lipids, and protein (Yumiko Ohkouchi & Inoue, 2007). In the fermentation of food residues with natural microbiota, it has been identified that BAL such as *L. plantarum* and *L. brevis* are the dominant microorganisms, which favors the production of LA (RedCorn & Engelberth, 2016). The fermentation of this type of waste in discontinuous systems with the addition of neutralizing agents generates a rapid hydrolysis of the raw material, and its main product is lactic acid (80%), as well as other by-products in lesser proportions (Gu, Liu, & Wong, 2018; Omar, Aini, Rahman, Hafid, & Yee, 2009). However, modifications of the culture system can affect the production of organic acids. Gu *et al.* (2018), determined that a semicontinuous culture, adding daily between 50 g/d and 150 g/d of organic load, drastically reduces the production of lactic acid at percentages lower than 10% and favors the production of butanoic acid. RedCorn (2016) determined that the optimal conditions of temperature, pH, fermentation time, and initial concentration of food residues, using natural microbiota as a catalyst, are 41 °C, pH 5.5, 20 h, and 150 g VS/L, respectively. Sakai *et al.* (2011), emphasized that food waste containing a high amount of total sugar close to 129 g/kg of waste can be transformed via enzymatic hydrolysis to glucose, reaching a concentration of 80 to 83 g/kg of waste. Kim *et al.* (2003), found a high LA yield of 0.91 g/g from coffee residue. From this residue, 80 g/L of LA were obtained in only 48 hours. Ohkouchi and Inoue (2006) also obtained a high yield of 1.11 g LA/g food waste, where the waste used showed a starch content of over 60% of total carbohydrates. The balance of nutrients and the amount of fermentable sugars in the culture medium is difficult to control when domestic wastes are used, since the chemical composition of these materials may vary from day to day. For that reason, Ohkouchi *et al.* (2007), determined that a 10:1 ratio of sugar/nitrogen must be supplied in order to generate high concentrations of LA. The study presented by Pleisener *et al.* (2017), on the manufacture of LA in a technical scale of 50 L under real process conditions, without sterilization of the medium, demonstrated the feasibility of the use of food waste for the industrial production of LA, with a starch product yield of 0.75 g/g and with no considerable production of other organic acids.

**Dairy industry.** Approximately 85% to 90% of all milk used for cheese production is discarded as whey, which

is mainly composed of lactose, proteins, and minerals. In Colombia, in 2009, for example, 1084 million liters of milk were used for cheese production, and 921.7 million liters were discarded as whey (Parra, 2009). There are two types of whey: sweet and acid. Both are successfully used in the production of LA, supplemented with yeast extract, glucose, peptone, soybean meal, or corn steep liquor. This sub-product is the most common substrate for the fermentative production of LA using LAB, and it has already been employed on an industrial scale by Sheffield By-Products (Hofvendahl & Hagerdal, 2000; Solá-Villatoro, 2006). *L. delbrueckii* spp. *bulgaricus*, *L. helveticus*, and *L. casei* are among the strains that have been used for the production of LA from whey. However, *L. helveticus* is preferred, because of its high rate of production (Panesar *et al.*, 2007). Urribarrí *et al.* (2004), determined that in a continuous culture of *L. helveticus* using deproteinized whey, the optimal conditions for the growth of this microorganisms are 40 °C, with pH 5,9 and a dilution rate of 2 h<sup>-1</sup>. The maximum specific growth rate was 0.47 h<sup>-1</sup>, and the specific rate of substrate consumption was 0.06 kg/m<sup>3</sup>. Volumetric productivity for the conversion of lactose to LA in continuous bioreactors has been reported to be about 2 to 4 g/Lh. Roukas and Kotzekidou (1998) found an increase of LA using fed batch fermentation and immobilized cells of *L. casei* and *Lactococcus lactis* with deproteinized whey as a substrate. Moreover, in the dairy industry, 1% of the total production of yogurt it is considered to be waste, with a high content of sucrose, glucose, and galactose, and it is more polluting than whey. Yogurt serum was used by Alonso *et al.* (2010), for the production of LA in a batch-type process, obtaining a yield of 90% and a maximum LA production of 25.90 g/L. The chemical oxygen demand (COD) was measured for yogurt serum before fermentation, showing a concentration of 90 g/L. After fermentation, it was reduced to 72.50 g/L. This demonstrates that the use of waste in fermentation processes helps to reduce the environmental impact. Table 3 shows a review of some economical raw materials that have been explored for LA fermentation.

Considering that some wastes are richer in nitrogen (e.g. wheat bran, corn cobs) and others as a source of carbon (e.g. starchy materials), depending on the availability of these raw materials in a region, the process could be optimized with reduced costs through a formulation that combines these wastes, reducing the purchase of commercial inputs. Less-developed regions such as Latin America and Africa produce millions of tons of the above waste annually, discarding it in most cases, without an integrated environmental plan. In spite of the commercial challenges that the generation of a new industry implies, the use of these raw materials for the production of LA in these regions of the planet is consid-



**Table 3.** Concentration, yield, production methods and productivity of L(+) - lactic acid using some agroindustrial and household wastes.

a.Prod. Productivity, b. Acid hydrolyses, c. Enzymatic hydrolyses, d. RFB - Rotating fibrous-bed bioreactor, e. BCR bubble column reactor.

Substrate	Microorganism	LA (g/L)	Yield (g/g)	Prod. <sup>a</sup> (g/L h)	Method	Reference
Alfalfa fibre	<i>L. delbrueckii</i>	35.40	0.35	0.75	SSF- Batch-47 h	(Greenath, Moldes, Koegel, & Straub, 2001)
Cassava bagasse	<i>L. delbrueckii</i> NCIM 2025	81.90	0.94	1.36	SSF- Batch-60 h	(John et al., 2006)
Coffee mucilage	<i>L. bulgaricus</i> NRRL-B548	41.00	1.20	1.44	SSF- Batch-25 h	(Arias Zabala, Henao Navarrete, & Castrillón Guitiérrez, 2009)
Corn cobs	<i>Rhizopus</i> sp. MK-96-1196	24.00	-	0.25	SSF- Batch -120 h	(Miura et al., 2004)
Corn cobs	<i>L. brevis</i>	39.10	0.75	0.82	AH <sup>b</sup> - Batch - 48 h	(Guo et al., 2010)
Corn starch	<i>R. oryzae</i> NRRL 395	127.00	1.00		EH <sup>c</sup> - RFB <sup>d</sup> - Fed Batch 90 h	(Tay & Yang, 2002)
Corn cobs	<i>R. oryzae</i> HZ56	48.80	0.79	0.87	AH <sup>b</sup> - Batch - 56 h	(Bai, Li, Liu, & Cui, 2008)
Food waste	<i>L. manihotivorans</i> LMC18011	48.70	-	1.11	SSF- Batch -140 h	(Y Ohkouchi & Inoue, 2006)
Pineapple waste	<i>L. casei</i> sub. <i>Rhamnosus</i>	75.00	0.98	3.90	SSF-Batch-40 h	(Araya-cloutier, Rojas-garbanzo, & Velázquez-carillo, 2010)
Waste potato starch	<i>R. arrhizus</i>	87.90	0.88	2.10	BCR <sup>e</sup> -Batch - 60 h	(Z. Y. Zhang, Jin, & Kelly, 2009)
Unpolished rice	<i>L. delbrueckii</i> HC106	90.80	0.73	1.50	SSF-Batch - 60 h	(Lu et al., 2009)
Oak wood chips	<i>Lactobacillus</i> sp. RKY2	42	0.95	6.70	Continuous-Mixture of pretreatment	(Y.-J. Wee & Ryu, 2009)
Cane molasses	<i>L. delbrueckii</i> mutante Uc-3	166	0.95	4.15	Batch-40 h	(Dumbrepail et al., 2008)

ered to present an opportunity for environmental and economic development.

### SIMULTANEOUS SACCHARIFICATION AND FERMENTATION OF LA

The bioconversion of materials to LA can be done in one step by coupling enzymatic hydrolysis and microbial fermentation. This process is known as simultaneous

saccharification and fermentation (SSF), (Marques et al., 2008; Watanabe et al., 2012). A benefit of SSF is the decrease in the inhibition by the substrate caused by the accumulation of sugars generated when enzymatic hydrolysis is performed separately, which allows an increase in the rate of saccharification and productivity. However, when cellulosic materials are used, inhibition by cellobiose remains even when SSF is implemented.

The addition of  $\beta$ -glucosidase at the beginning of SSF could help to avoid this problem (M. Adsul, Khire, Bastawde, & Gokhale, 2007; John *et al.*, 2007; Ping *et al.*, 2005). SSF can be successfully carried out when the microorganisms and enzyme systems have similar culture conditions, because if there is a marked difference between them, the production of LA is drastically affected. For that reason, identification of the microbial and biochemical kinetics is of great importance for determining the optimum conditions for improving the performance of SSF (John *et al.*, 2006; Venkatesh, 1997). Marques *et al.* (2008), conducted a comparison of separate and SSF using paper industry waste. They obtained better results with fermentation in a single step, and the production of LA increased from 51.90 g/L to 72.90 g/L. As mentioned above, the use of commercial enzymes for the hydrolysis process generates a considerable expense for the manufacture of LA. Due to this, other types of cultures have been explored. For example, microorganisms produce LA capable of excreting enzymes to carry out the hydrolysis, and the mixture of different species of microorganisms (co-cultures) for the development of SSF increases the economic feasibility of the whole process (Pleissner *et al.*, 2017; Watanabe *et al.*, 2012). Miura *et al.* (2004), developed a mixed culture for the production of LA using corn cob as a substrate; *A. thermophilus* (producer of cellulase) was in charge of enzymatic hydrolysis, while *Rhizopus sp.* was used for the production of LA. The same co-culture strategy was used by John *et al.*, (2007). They mixed two strains of *Lactobacillus* in order to degrade cassava bagasse: *L. casei* and *L. delbrueckii*. Aerobic microorganisms such as *Aspergillus sp.* are used in the industry to produce amylase. Kurosawa *et al.* (1988), performed an immobilization of a mixture of *Aspergillus awamori* for saccharification and *Streptococcus lactis* (aerotolerant anaerobic) for fermentation with starchy substrates. They generated 25 g/L LA from 50 g/L of substrate.

#### GENETIC STRATEGIES FOR INCREASING THE PRODUCTION OF LA

The metabolism of LA microorganisms has been modified using genetic and metabolic engineering tools in order to increase their production of LA or to introduce exogenous genes in order to convert the microorganisms into producers of this metabolite. *Rhizopus* strains have two lactate dehydrogenase (LDH) genes, *ldhA* and *ldhB*. *LdhA* is believed to be primarily responsible for the production of LA (John *et al.*, 2009; Christopher D. Skory, Mertens, & Rich, 2009). That is why Skory (2004) tried to increase LDH activity by determining the length of the *ldhA* gene and increasing the number of copies of this gene in *R. oryzae*. This resulted in an increase of specific LDH activity and LA production. Furthermore, there

was a decrease in by-products when compared with the control case. In other studies, *Rhizopus sp.* was transformed in order to decrease the activity of the alcohol dehydrogenase enzyme, obtaining a mutant strain capable of generating 35% less ethanol and about ten times more LA than the parent strain (Christopher D Skory, Freer, & Bothast, 1998). Moreover, *L. delbrueckii* UC-3 is a strain mutated by exposure to ultraviolet light. It has the ability to tolerate high sugar concentrations, making it an attractive microorganism for LA fermentation with molasses waste and sugarcane bagasse. This microorganism also has enzymes that can degrade cellobiose and could be used with this type of substrate (M. Adsul *et al.*, 2007; Dumbrepatil *et al.*, 2008). Zhou *et al.* (2011), obtained a recombinant strain with multiple deletions (*E. coli* CICIM B013-070) able to produce 125 g LA/L from glucose. Its low requirements for minerals and its microaerobic conditions make this strain suitable for large-scale use. Yeast does not have the *Ldh* enzyme, and therefore it lacks the ability to produce LA. Ishida *et al.* (2005), reported the successful transformation of *Saccharomyces cerevisiae* by the insertion of two copies of the LDH gene on the genome, which express bovine LDH under the control of native promoter pyruvate decarboxylase1 (*PDC1*). This strain proved to be an efficient and pH-tolerant microorganism, which produced approximately 50 g/L LA(+) with high optical purity without the need for neutralization (M.-T. Gao, Shimamura, Ishida, & Takahashi, 2009). The problem presented by *S. cerevisiae* is the large amount of ethanol produced. *Kluyveromyces lactis* has only one gene that codifies for pyruvate decarboxylase, while *S. cerevisiae* has three genes for this enzyme. Sinhg *et al.* (2006), suppressed this gene in *K. lactis* and avoided the production of ethanol. Thus only LA was generated, with a high concentration of 109 g/L. In more recent studies, Lee *et al.* (2017), obtained *K. marxianus* yeast able to co-express two foreign *Ldh* enzymes, each with a different optimum pH (basic or acid), and the production of LA improved. These results show the importance of a pH drop in the culture and its effect on the production of lactic acid.

#### CONCLUSIONS

The high content of cellulose, hemicellulose, starch, and lactose in agroindustrial and household wastes make them a replacement for expensive pure sugars as carbon sources. In some cases, they can also be utilized as a nutrient source for LA production by fermentation. However, the use of these materials in LA production has some drawbacks, due to the pretreatment required. SSF and the use of recombinant microorganisms have made a great contribution to the increase of LA production, overcoming some problems, such as high content of

undesirable by-products and inhibition by sugars in the media. In the pre-treatment of the wastes, enzymatic hydrolysis is considered to be more beneficial for the performance of the microorganism, this in spite of the higher cost of this treatment. Although both fungi and LAB are reported to give very good yields in the production of LA using these residues, LAB are considered more promising, due to their operational advantages and the greater facility of genetic transformation, which increases production. In spite of the commercial challenges that the generation of a new industry implies, the use of these raw materials for the production of LA in regions as Latin America and Africa is considered to be an opportunity for environmental and economic development.

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## BIBLIOGRAPHY

- Abdel-Rahman, M. A., Tashiro, Y., & Sonomoto, K. (2011). Lactic acid production from lignocellulose-derived sugars using lactic acid bacteria: Overview and limits. *Journal of Biotechnology*, 156(4), 286–301. <http://doi.org/10.1016/j.jbiotec.2011.06.017>.
- Abdel-Rahman, M. A., Tashiro, Y., & Sonomoto, K. (2013). Recent advances in lactic acid production by microbial fermentation processes. *Biotechnology Advances*, 6–10. <http://doi.org/10.1016/j.biotechadv.2013.04.002>.
- Adsul, M. G., Varma, A. J., & Gokhale, D. V. (2007). Lactic acid production from waste sugarcane bagasse derived cellulose. *Green Chem*, 58–62.
- Adsul, M., Khire, J., Bastawde, K., & Gokhale, D. (2007). Production of lactic acid from cellobiose and celotriose by *Lactobacillus delbrueckii* mutant Uc-3. *Applied and Environmental Microbiology*, 73(15). <http://doi.org/10.1128/AEM.00774-07>.
- Alonso, S., Herrero, M., Rendueles, M., & Díaz, M. (2010). Residual yoghurt whey for lactic acid production. *Biomass and Bioenergy*, 34(7), 931–938. <http://doi.org/10.1016/j.biombioe.2010.01.041>.
- Araya-cloutier, C., Rojas-garbanzo, C., & Velázquez-carillo, C. (2010). Síntesis de ácido láctico, a través de la hidrólisis enzimática simultánea a la fermentación de un medio a base de un desecho de piña (*Ananas comosus*), para su uso como materia prima en la elaboración de ácido poliláctico. *Revista Iberoamericana de Polímeros*, 11(7), 407–416.
- Arias Zabala, M., Henao Navarrete, L., & Castrillón Guitiérrez, Y. (2009). Producción de ácido láctico por fermentación de mucílago de café con *Lactobacillus bulgaricus* NRRL-B548. *Dyna*, 76 (158), 147–153.
- Asocaña. (2012). El sector azucarero en la actualidad. Retrieved January 1, 2013, from [www.asocana.org](http://www.asocana.org).
- Bai, D.-M., Li, S.-Z., Liu, Z. L., & Cui, Z.-F. (2008). Enhanced L (+) lactic acid production by an adapted strain of *Rhizopus oryzae* using corncob hydrolysate. *Applied Biochemistry and Biotechnology*, 79–85.
- Bilanovic, D., Chang, F., Isobaev, P., & Welle, P. (2011). Lactic acid and xanthan fermentations on an alternative potato residues media - Carbon source costs. *Biomass and Bioenergy*, 35(7), 2683–2689. <http://doi.org/10.1016/j.biombioe.2011.03.001>.
- Bishai, M., De, S., Adhikari, B., & Banerjee, R. (2013). *Zizyphus oenoplia*: a potent substrate for lactic acid production. *Bioresource Technology*, 133, 627–629. <http://doi.org/10.1016/j.biortech.2012.12.049>.
- Boonpan, A., Pivsa-art, S., Pongswat, S., Areesirisuk, A., & Sirisangsawang, P. (2013). Separation of D, L-Lactic Acid by Filtration Process. *Energy Procedia*, 34 (662), 898–904. <http://doi.org/10.1016/j.egypro.2013.06.827>.
- Bulut, S., Elibol, M., & Ozer, D. (2004). Effect of different carbon sources on L (+) -lactic acid production by *Rhizopus oryzae*. *Biochemical Engineering Journal*, 21, 33–37. <http://doi.org/10.1016/j.bej.2004.04.006>.
- Castaño, H., & Mejía, C. (2008). Producción de etanol a partir de almidón de yuca utilizando la estrategia de proceso sacarificación- fermentación simultáneas (SSF). *Vitae*, 15, 251–258.
- Castillo Martínez, F. A., Balciunas, E. M., Salgado, J. M., Domínguez González, J. M., Converti, A., & Oliveira, R. P. D. S. (2013). Lactic acid properties, applications and production: A review. *Trends in Food Science & Technology*, 30(1), 70–83. <http://doi.org/10.1016/j.tifs.2012.11.007>.
- Cock, L., & Rodríguez, A. (2005). Producción Biotecnológica de Ácido Láctico: Estado del arte. *Ciencia y Tecnología Alimentaria*, 5, 54–65.
- Corbion. (2019). Corbion. Retrieved from <https://www.corbion.com/>
- Cuervo, L., Folch, J. L., & Quiroz, R. E. (2001). Lignocelulosa Como Fuente de Azúcares Para la Producción de Etanol. *BioTecnología*, 13(3), 11–25.
- Doherty, W. O. S., Mousavioun, P., & Fellows, C. M. (2011). Value-adding to cellulosic ethanol: Lignin polymers. *Industrial Crops and Products*, 33(2), 259–276. <http://doi.org/10.1016/j.indcrop.2010.10.022>.
- Domínguez, M., Castillo, A. Á., Castrejón, T., Granados,

- M., Hernandez, F., Alcalá, V. H., & Picazo, J. (2011). Estudio de la cinética de la hidrólisis áida del bgazo de caña de azúcar sin pretratamiento para la obtención de azúcares reductores. *Revista Iberoamericana de Polímeros*, 12(3), 153–159.
- Dumbrepatil, A., Adsul, M., Chaudhari, S., Khire, J., & Gokhale, D. (2008). Utilization of molasses sugar for lactic acid production by *Lactobacillus delbrueckii* subsp. *delbrueckii* mutant Uc-3 in batch fermentation. *Applied and Environmental Microbiology*, 74(1), 333–335. <http://doi.org/10.1128/AEM.01595-07>.
- Galactic. (2019). Galactic. Retrieved from <https://www.lactic.com/en-us/home.aspx>.
- Gao, C., Ma, C., & Xu, P. (2011). Biotechnological routes based on lactic acid production from biomass. *Biotechnology Advances*, 29(6), 930–939. <http://doi.org/10.1016/j.biotechadv.2011.07.022>.
- Gao, L., Yang, H., Wang, X., Huang, Z., Ishii, M., Igarashi, Y., & Cui, Z. (2008). Rice straw fermentation using lactic acid bacteria. *Bioresource Technology*, 99, 2742–2748. <http://doi.org/10.1016/j.biortech.2007.07.001>.
- Gao, M.-T., Shimamura, T., Ishida, N., & Takahashi, H. (2009). Application of metabolically engineered *Saccharomyces cerevisiae* to extractive lactic acid fermentation. *Biochemical Engineering Journal*, 44(2–3), 251–255. <http://doi.org/10.1016/j.bej.2009.01.001>.
- Gao, M., Kaneko, M., Hirata, M., Toorisaka, E., & Hano, T. (2008). Utilization of rice bran as nutrient source for fermentative lactic acid production. *Bioresource Technology*, 99, 3659–3664. <http://doi.org/10.1016/j.biortech.2007.07.025>.
- Gu, X. Y., Liu, J. Z., & Wong, J. W. C. (2018). Control of lactic acid production during hydrolysis and acidogenesis of food waste. *Bioresource Technology*, 247(July 2017), 711–715. <http://doi.org/10.1016/j.biortech.2017.09.166>.
- Guo, W., Jia, W., Li, Y., & Chen, S. (2010). Performances of *Lactobacillus brevis* for producing lactic acid from hydrolysate of lignocellulosics. *Applied Biochemistry and Biotechnology*, 161(1–8), 124–136. <http://doi.org/10.1007/s12010-009-8857-8>.
- Hofvendahl, K., & Hagerdal, B. H. (2000). Factors affecting the fermentative lactic acid production from renewable resources. *Enzyme and Microbial Technology*, 26, 87–107.
- Howard, R. L., Abotsi, E., L, J. V. R. E., & Howard, S. (2003). Lignocellulose biotechnology: issues of bioconversion and enzyme production. *African Journal of Biotechnology*, 2(December), 602–619.
- Hu, Y., Kwan, T. H., Daoud, W. A., Sze, C., & Lin, K. (2017). Continuous ultrasonic-mediated solvent extraction of lactic acid from fermentation broths. *Journal of Cleaner Production*, 145, 142–150. <http://doi.org/10.1016/j.jclepro.2017.01.055>.
- Huang, L. P., Jin, B., & Lant, P. (2005). Direct fermentation of potato starch wastewater to lactic acid by *Rhizopus oryzae* and *Rhizopus arrhizus*. *Bioprocess and Biosystems Engineering*, 27(4), 229–38. <http://doi.org/10.1007/s00449-005-0398-0>.
- Ishida, N., Saitoh, S., Tokuhira, K., Nagamori, E., Matsuyama, T., Kitamoto, K., & Takahashi, H. (2005). Efficient Production of L-Lactic Acid by Metabolically Engineered *Saccharomyces cerevisiae* with a Genome-Integrated L-Lactate Dehydrogenase Gene. *Applied and Environmental Microbiology*, 71(4), 1964–1970. <http://doi.org/10.1128/AEM.71.4.1964>.
- John, R. P., Anisha, G. S., Nampoothiri, K. M., & Pandey, A. (2009). Direct lactic acid fermentation: Focus on simultaneous saccharification and lactic acid production. *Biotechnology Advances*, 27(2), 145–152. <http://doi.org/10.1016/j.biotechadv.2008.10.004>.
- John, R. P., Nampoothiri, K. M., & Pandey, A. (2006). Simultaneous Saccharification and Fermentation of Cassava Bagasse for L-(+)-Lactic Acid Production Using *Lactobacilli*. *Applied Biochemistry and Biotechnology*, 134, 263–272.
- John, R. P., Sukumaran, R. K., Nampoothiri, K. M., & Pandey, A. (2007). Statistical optimization of simultaneous saccharification and L-(+)-lactic acid fermentation from cassava bagasse using mixed culture of *Lactobacilli* by response surface methodology. *Biochemical Engineering Journal*, 36, 262–267. <http://doi.org/10.1016/j.bej.2007.02.028>.
- Kurosawa, H., Ishikawa, H., & Tanaka, H. (1988). L-Lactic Acid Production from Starch by Coimmobilized Mixed Culture System of *Aspergillus awamori* and *Streptococcus lactis*. *Biotechnology and Bioengineering*, 31, 183–187.
- Laopaiboon, P., Thani, A., Leelavatcharamas, V., & Laopaiboon, L. (2010). Acid hydrolysis of sugarcane bagasse for lactic acid production. *Bioresource Technology*, 101(3), 1036–1043. <http://doi.org/10.1016/j.biortech.2009.08.091>.
- Lee, J. W., In, J. H., Park, J. B., Shin, J., Park, J. H., Sung, B. H., ... Kweon, D. H. (2017). Co-expression of two heterologous lactate dehydrogenases genes in *Kluyveromyces marxianus* for L-lactic acid production. *Journal of Biotechnology*, 241, 81–86. <http://doi.org/10.1016/j.jbiotec.2016.11.015>.
- Lee, K. (2005). A media design program for lactic acid production coupled with extraction by electro dialysis. *Bioresource Technology*, 96(13), 1505–1510. <http://doi.org/10.1016/j.biortech.2004.11.010>.

- Li, Z., Han, L., Ji, Y., Wang, X., & Tan, T. (2010). Fermentative production of L-lactic acid from hydrolysate of wheat bran by *Lactobacillus rhamnosus*. *Biochemical Engineering Journal*, 49(1), 138–142. <http://doi.org/10.1016/j.bej.2009.10.014>.
- Li, Z., Tianwei, T., Jike, L., Zixin, Y., & Lu, H. (2012). Utilization of White rice bran for production of L-lactic acid. *Biomass and bioenergy*. *Biomass and Bioenergy*, 1–6.
- Liu, Y., Liao, W., Liu, C., & Chen, S. (2006). Optimization of L-(+)-Lactic Acid Production Using Pelletized Filamentous *Rhizopus oryzae* NRRL 395. *Applied Biochemistry and Biotechnology*, 129, 844–853.
- Lu, Z., Lu, M., He, F., & Yu, L. (2009). An economical approach for d-lactic acid production utilizing unpolished rice from aging paddy as major nutrient source. *Bioresource Technology*, 100(6), 2026–2031. <http://doi.org/10.1016/j.biortech.2008.10.015>.
- Maas, R. H. W., Bakker, R. R., Eggink, G., & Weusthuis, R. a. (2006). Lactic acid production from xylose by the fungus *Rhizopus oryzae*. *Applied Microbiology and Biotechnology*, 72(5), 861–8. <http://doi.org/10.1007/s00253-006-0379-5>.
- Maria, D., & Valencia, A. (2011). Producción de etanol a partir de bagazo de caña panelera mediante un sistema híbrido de fermentación y pervaporación.
- María, S., & Echeverri, P. (2004). Los residuos sólidos municipales como acondicionadores de suelos. *Revista La Sallista de Investigación*, 1, 56–65.
- Marques, S., Santos, A. L., Francisco, M. G., & Roseiro, J. C. (2008). Lactic acid production from recycled paper sludge by simultaneous saccharification and fermentation. *Biochemical Engineering Journal*, 41, 210–216. <http://doi.org/10.1016/j.bej.2008.04.018>.
- Merino, S. T., & Cherry, J. (2007). Progress and challenges in enzyme development for biomass utilization. *Advances in Biochemical Engineering/Biotechnology*, 108(June), 95–120. [http://doi.org/10.1007/10\\_2007\\_066](http://doi.org/10.1007/10_2007_066).
- Miura, S., Arimura, T., Itoda, N., Dwiarti, L., Feng, J. I. N., Bin, C. U. I. H., & Okabe, M. (2004). Production of L-Lactic Acid from Corn cob. *Journal of Bioscience and Bioengineering*, 97(3), 153–157.
- Nakano, S., Ugwu, C. U., & Tokiwa, Y. (2012). Bioresource Technology Efficient production of D-(–)-lactic acid from broken rice by *Lactobacillus delbrueckii* using Ca(OH)<sub>2</sub> as a neutralizing agent. *Bioresource Technology*, 104, 791–794. <http://doi.org/10.1016/j.biortech.2011.10.017>.
- Natureworks LLC. (2019). <https://www.natureworksllc.com/What-is-Ingeo/How-Ingeo-is-Made>.
- Nuttha, T. (2005). Lactic acid production by immobilized *Rhizopus oryzae* in a rotating fibrous bed bioreactor.
- Oda, Y., Saito, K., Yamauchi, H., & Mori, M. (2002). Lactic acid fermentation of potato pulp by the fungus *Rhizopus oryzae*. *Current Microbiology*, 45(1), 1–4. <http://doi.org/10.1007/s00284-001-0048-y>.
- Oh, H., Wee, Y.-J., Yun, J.-S., Ho Han, S., Jung, S., & Ryu, H.-W. (2005). Lactic acid production from agricultural resources as cheap raw materials. *Bioresource Technology*, 96(13), 1492–1498. <http://doi.org/10.1016/j.biortech.2004.11.020>.
- Ohkouchi, Y., & Inoue, Y. (2006). Direct production of L+lactic acid from starch and food wastes using *Lactobacillus manihotivorans* LMG18011. *Bioresource Technology*, 97(13), 1554–1562. <http://doi.org/10.1016/j.biortech.2005.06.004>.
- Ohkouchi, Y., & Inoue, Y. (2007). Impact of chemical components of organic wastes on L(+)-lactic acid production. *Bioresource Technology*, 98(3), 546–53. <http://doi.org/10.1016/j.biortech.2006.02.005>.
- Omar, F. N., Aini, N., Rahman, A., Hafid, H. S., & Yee, P. L. (2009). Separation and recovery of organic acids from fermented kitchen waste by an integrated process. *African Journal of Biotechnology*, 8(21), 5807–5813.
- Pal, P., Sikder, J., Roy, S., & Giorno, L. (2009). Process intensification in lactic acid production: A review of membrane based processes. *Chemical Engineering and Processing*, 48, 1549–1559. <http://doi.org/10.1016/j.cep.2009.09.003>.
- Panesar, P. S., Kennedy, J. F., Gandhi, D. N., & Bunko, K. (2007). Food Chemistry Bioutilisation of whey for lactic acid production. *Food Chemistry*, 105, 1–14. <http://doi.org/10.1016/j.foodchem.2007.03.035>.
- Parra, R. (2009). Lactosuero: importancia en la industria de alimentos. *Revista Facultad Nacional de Agronomía*, 62(1), 4967–4982.
- Ping, L., Jin, B., Lant, P., & Zhou, J. (2005). Simultaneous saccharification and fermentation of potato starch wastewater to lactic acid by *Rhizopus oryzae* and *Rhizopus arrhizus*. *Biochemical Engineering Journal*, 23, 265–276. <http://doi.org/10.1016/j.bej.2005.01.009>.
- Pleissner, D., Demichelis, F., Mariano, S., Fiore, S., Navarro Gutiérrez, I. M., Schneider, R., & Venus, J. (2017). Direct production of lactic acid based on simultaneous saccharification and fermentation of mixed restaurant food waste. *Journal of Cleaner Production*, 143, 615–623. <http://doi.org/10.1016/j.jclepro.2016.12.065>.
- RedCorn, R., & Engelberth, A. S. (2016). Identifying

- conditions to optimize lactic acid production from food waste co-digested with primary sludge. *Biochemical Engineering Journal*, 105, 205–213. <http://doi.org/10.1016/j.bej.2015.09.014>.
- RedCorn, R., Fatemi, S., & Engelberth, A. S. (2018). Comparing End-Use Potential for Industrial Food-Waste Sources. *Engineering*, 4(3), 371–380. <http://doi.org/10.1016/j.eng.2018.05.010>.
- Reddy, G., Altaf, M., Naveena, B. J., Venkateshwar, M., & Kumar, E. V. (2008). Amylolytic bacterial lactic acid fermentation - a review. *Biotechnology Advances*, 26(1), 22–34. <http://doi.org/10.1016/j.biotechadv.2007.07.004>.
- Rojan, P. J., Nampoothiri, K. M., Nair, A. S., & Pandey, A. (2005). L(+)-lactic acid production using *Lactobacillus casei* in solid-state fermentation. *Biotechnology Letters*, 27(21), 1685–1688. <http://doi.org/10.1007/s10529-005-2731-8>.
- Roukas, T., & Kotzekidou, P. (1998). Lactic acid production from deproteinized whey by mixed cultures of free and coimmobilized *Lactobacillus casei* and *Lactococcus lactis* cells using fedbatch culture. *Enzyme and Microbial Technology*, 22(97), 199–204.
- Ruengruglikit, C., & Hang, Y. D. (2003). L(+)-Lactic acid production from corncobs by *Rhizopus oryzae* NRRL-395. *LWT - Food Science and Technology*, 36(6), 573–575. [http://doi.org/10.1016/S0023-6438\(03\)00062-8](http://doi.org/10.1016/S0023-6438(03)00062-8).
- Saito, K., Hasa, Y., & Abe, H. (2012). Production of lactic acid from xylose and wheat straw by *Rhizopus oryzae*. *Journal of Bioscience and Bioengineering*, 114(2), 166–169. <http://doi.org/10.1016/j.jbiosc.2012.03.007>.
- Sakai, K., Poudel, P., & Shirai, Y. (2011). Total Recycle System of Food Waste for Poly-L-Lactic Acid Output. *Advances in Applied Biotechnology*, 2.
- Serna, L., & Rodríguez, A. (2007). Producción económica de ácido láctico utilizando residuos de cosecha y jugos de caña de azúcar (*Saccharum officinarum* L.). *Agricultura Técnica*, 67(1), 29–38.
- Shibata, K., Flores, D. M., Kobayashi, G., & Sonomoto, K. (2007). Direct l-lactic acid fermentation with sago starch by a novel amylolytic lactic acid bacterium, *Enterococcus faecium*. *Enzyme and Microbial Technology*, 41(1–2), 149–155. <http://doi.org/10.1016/j.enzmictec.2006.12.020>.
- Singh, S. K., Ahmed, S. U., & Pandey, A. (2006). Metabolic engineering approaches for lactic acid production. *Process Biochemistry*, 41, 991–1000. <http://doi.org/10.1016/j.procbio.2005.12.004>.
- Skory, C. D. (2004). Lactic acid production by *Rhizopus oryzae* transformants with modified lactate dehydrogenase activity. *Applied Microbiology and Biotechnology*, 64(2), 237–242. <http://doi.org/10.1007/s00253-003-1480-7>.
- Skory, C. D., Freer, S. N., & Bothast, R. J. (1998). Production of L-lactic acid by *Rhizopus oryzae* under oxygen limiting conditions. *Biotechnology Letters*, 20(2), 191–194.
- Skory, C. D., Mertens, J. a., & Rich, J. O. (2009). Inhibition of *Rhizopus lactate dehydrogenase* by fructose 1,6-bisphosphate. *Enzyme and Microbial Technology*, 44(4), 242–247. <http://doi.org/10.1016/j.enzmictec.2008.10.009>.
- Solá-Villatoro, G. (2006). Estudio de factibilidad para la producción de ácido láctico comercial, a nivel industrial en Guatemala.
- Sreenath, H. K., Moldes, a B., Koegel, R. G., & Straub, R. J. (2001). Lactic acid production by simultaneous saccharification and fermentation of alfalfa fiber. *Journal of Bioscience and Bioengineering*, 92(6), 518–523. Retrieved from <http://www.ncbi.nlm.nih.gov/pubmed/16233139>.
- Sun, Y., & Cheng, J. (2002). Hydrolysis of lignocellulosic materials for ethanol production: a review. *Bioresource Technology*, 83(1), 1–11. Retrieved from <http://www.ncbi.nlm.nih.gov/pubmed/12058826>.
- Tay, A., & Yang, S. (2002). Production of L(+)-lactic acid from glucose and starch by immobilized cells of *Rhizopus oryzae* in a rotating fibrous bed bioreactor. *Biotechnology Bioeng*, 80, 1–12.
- Teixeira, E. D. M., Curvelo, A. A. S., Corrêa, A. C., Marconcini, J. M., Glenn, G. M., & Mattoso, L. H. C. (2012). Properties of thermoplastic starch from cassava bagasse and cassava starch and their blends with poly ( lactic acid ). *Industrial Crops & Products*, 37(1), 61–68. <http://doi.org/10.1016/j.indcrop.2011.11.036>.
- Urribarrí, L., Vielma, A., Paéz, G., Ferrer, J., & Mármol, Z. (2004). Producción de ácido láctico a partir de suero de leche , utilizando *Lactobacillus helveticus* en cultivo continuo. *Redalyc*.
- Van Dyk, J. S., & Pletschke, B. I. (2012). A review of lignocellulose bioconversion using enzymatic hydrolysis and synergistic cooperation between enzymes–factors affecting enzymes, conversion and synergy. *Biotechnology Advances*, 30(6), 1458–80. <http://doi.org/10.1016/j.biotechadv.2012.03.002>.
- Venkatesh, K. V. (1997). Simultaneous saccharification and fermentation of cellulose to lactic acid. *Biotechnology and Bioengineering*, 62(1), 91–98. <http://doi.org/10.1002/bit.260370113>.
- Vishnu, C., Seenayya, G., & Reddy, G. (2002). Direct fermentation of various pure and crude starchy substrates to L ( + ) lactic acid using *Lactobacillus amylophilus* GV6. *World Journal of Microbiology*



- and *Biotechnology*, 18, 429–433.
- Wang, Z., Wang, Y., Yang, S.-T., Wang, R., & Ren, H. (2010). A novel honeycomb matrix for cell immobilization to enhance lactic acid production by *Rhizopus oryzae*. *Bioresource Technology*, 101(14), 5557–5564. <http://doi.org/10.1016/j.biortech.2010.02.064>.
- Watanabe, M., Makino, M., Kaku, N., Koyama, M., Nakamura, K., & Sasano, K. (2012). Fermentative L - ( + ) -lactic acid production from non-sterilized rice washing drainage containing rice bran by a newly isolated lactic acid bacteria without any additions of nutrients. *Journal of Bioscience and Bioengineering*, 1–4. <http://doi.org/10.1016/j.jbiosc.2012.11.001>.
- Watanabe, M., Techapun, C., Kuntiya, A., Leksawasdi, N., Seesuriyachan, P., Chaiyaso, T., ... Nakamura, K. (2016). Extracellular protease derived from lactic acid bacteria stimulates the fermentative lactic acid production from the by-products of rice as a biomass refinery function. *Journal of Bioscience and Bioengineering*, 123(2), 245–251. <http://doi.org/10.1016/j.jbiosc.2016.08.011>.
- Wee, Y.-J., Kim, J.-N., Yun, J.-S., & Ryu, H.-W. (2004). Utilization of sugar molasses for economical l(+)-lactic acid production by batch fermentation of *Enterococcus faecalis*. *Enzyme and Microbial Technology*, 35(6–7), 568–573. <http://doi.org/10.1016/j.enzmictec.2004.08.008>.
- Wee, Y.-J., & Ryu, H.-W. (2009). Lactic acid production by *Lactobacillus* sp. RKY2 in a cell-recycle continuous fermentation using lignocellulosic hydrolyzates as inexpensive raw materials. *Bioresource Technology*, 100(18), 4262–4270. <http://doi.org/10.1016/j.biortech.2009.03.074>
- Wee, Y., Kim, J., & Ryu, H. (2006). Biotechnological Production of Lactic Acid and Its Recent Applications. *Food Technol Biotech*, 44, 163–172.
- Woiciechowski, A. L., Nitsche, S., Pandey, A., & Ricardo, C. (2002). Acid and enzymatic Hydrolysis to Recover Reducing Sugars from Cassava Bagasse:an Economic Study. *Brazilian Archives of Biologu and Technology*, 45, 393–400.
- Woiciechowski, A. L., Soccol, C. R., Ramos, L. P., & Pandey, A. (1999). Experimental design to enhance the production of L - ( + ) -lactic acid from steam-exploded wood hydrolysate using *Rhizopus oryzae* in a mixed-acid fermentation. *Process Biochemistry*, 34, 949–955.
- Yousuf, A., Bastidas-Oyanedel, J. R., & Schmidt, J. E. (2018). Effect of total solid content and pretreatment on the production of lactic acid from mixed culture dark fermentation of food waste. *Waste Management*, 77, 516–521. <http://doi.org/10.1016/j.wasman.2018.04.035>.
- Zhang, Z. (2008). Optimisation and Scale-up of a Biotechnological Process for Production of L ( + ) -Lactic Acid from Waste Potato Starch by *Rhizopus arrhizus*.
- Zhang, Z. Y., Jin, B., & Kelly, J. M. (2007). Production of lactic acid from renewable materials by *Rhizopus* fungi. *Biochemical Engineering Journal*, 35, 251–263. <http://doi.org/10.1016/j.bej.2007.01.028>.
- Zhang, Z. Y., Jin, B., & Kelly, J. M. (2009). Enhancement of L ( + ) -lactic acid production using acid-adapted precultures of *Rhizopus arrhizus* in a bubble column reactor. *JBIOSC*, 108(4), 344–347. <http://doi.org/10.1016/j.jbiosc.2009.04.009>.
- Zhou, L., Zuo, Z.-R., Chen, X.-Z., Niu, D.-D., Tian, K.-M., Prior, B. a, ... Wang, Z.-X. (2011). Evaluation of genetic manipulation strategies on D-lactate production by *Escherichia coli*. *Current Microbiology*, 62(3), 981–989. <http://doi.org/10.1007/s00284-010-9817-9>.