

Volatile Organic Constituents of *Clinopodium gilliesii* (Benth.) Kuntze: Analysis by HS-SPME and classic hydrodistillation

Constituyentes Orgánicos Volátiles de *Clinopodium gilliesii* (Benth.) Kuntze: Análisis por HS-SPME e hidrodestilación clásica

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Abstract

In the present research, an analytical methodology to micro scale based on the use of the HS-SPME/GC-MS to determine volatile compounds present in *Clinopodium gilliesii* (Benth.) Kuntze (Lamiaceae) was employed, and settled differences and similarities with its essential oil obtained by hydrodistillation. A systematic description of the volatile components of flowers, stems, leaves and combined aerial parts (whole plant) was constructed via GC-MS analyses of HS-SPME adsorbed compounds and of essential oils obtained through hydrodistillation of the same tissues. Piperitenone oxide and piperitone oxide were the main components of both the HS-SPME analysis and essential oil analysis. The HS-SPME method can achieve comparable results to those obtained by essential oil analysis, by using very fewer samples, a shorter extraction time and a much simpler procedure.

Keywords: Essential oil, piperitenone oxide, piperitone oxide, volatile organic compounds.

Resumen

En el presente trabajo se empleó una metodología analítica a micro-escala basada en HS-SPME/GC-MS, para determinar los compuestos volátiles presentes en *Clinopodium gilliesii* (Benth.) Kuntze (Lamiaceae), y se establecieron diferencias y similitudes con su aceite esencial obtenido por hidrodestilación. Se realizó una descripción sistemática de los componentes volátiles de flores, tallos, hojas y partes aéreas combinadas (planta entera) a partir de los análisis por GC-MS, a través del sistema HS-SPME y de los aceites esenciales. La piperitenona y el óxido de piperitona fueron los componentes principales tanto del análisis por HS-SPME, como del aceite esencial. El método de HS-SPME puede lograr resultados comparables a los obtenidos por el análisis de aceite esencial, mediante el uso de muestras de menor tamaño, un tiempo de extracción más corto y un procedimiento más simple.

Palabras clave: Aceite esencial, compuestos orgánicos volátiles, óxido de piperitenona, óxido de piperitona.

INTRODUCTION

Clinopodium gilliesii (Benth.) Kuntze (Lamiaceae, synonymy *Satureja parvifolia* (Philippi) Epling, figure 1) is an aromatic plant that grows in Perú, Chile, Bolivia and at the verge of rivers descending from the hills in the northwestern provinces of Argentina (Salta, Jujuy, Córdoba, Catamarca, Tucumán, La Rioja, San Juan, San Luis). In our country it is known with the common name of “muña muña” (Barbosa *et al.*, 2006) and is popularly used for its aphrodisiac properties (Viturro *et al.*, 2007). Usually found in health food local stores marketed as vegetable crude drugs and its production comes entirely from wild collection. So that, this species supports a high extraction pressure resulting from its use in herbalism, yerba mate composite (traditional infusion).

The fresh or dried herb is used as a flavoring agent for aliments in cooking, and in traditional medicine of the people of the mountains, as aphrodisiac, digestive, stimulant, emmenagogue, against altitude sickness, colds and female sterility (Viturro *et al.*, 2007).

Moreover, investigations on this plant deal with the chemical composition and the antifungal activity of its essential oil (Viturro *et al.*, 2007). Hnatyszyn *et al.* (2003), has reported a very interesting study related to ethnopharmacological use of this plant as aphrodisiac. In this work a systematic study of different extracts of *C. gilliesii* and other plants over relaxant activity in the smooth muscle of the corpus cavernosum on the Guinea pig was performed. As a result, the dichloromethane extract of *C. gilliesii* was very active in at a very low dose (2.5 mg.ml⁻¹). Thus these results reported here give some scientific supports to one of the more traditional uses as aphrodisiac of this plant.

Additionally, other biological activities for this plant include, insect-repellent, antifungal and antibacterial (Oliveira *et al.*, 2011).

Volatile Organic Constituents (VOCs) are defined as any organic compound with a boiling point in the range from (50° to 100°C) to (240° to 260°C), corresponding to having saturation vapour pressures at 25°C greater than 100 kPa. Studies on volatile composition of the essential oil obtained by hydrodistillation of aerial parts of *C. gilliesii* have indicated that piperitone oxide was the majoritary component. Additionally, pulegone, limonene, dodecanal, citronellol, sphaulenol, α -terpineol and menthone, were also present, although in smaller amounts. However, has also been reported that piperitenone oxide was the largest component and other majority component but in minor proportion were

piperitenone and pulegone (Viturro *et al.*, 2007). These differences in the composition of volatile compounds can be revealing the existence of chemotypes within this species.

Several techniques have been used to characterize volatile organic compounds present in plants, although hydrodistillation is the most common extraction technique employed to obtain essential oils from aromatic plants (Saroglou *et al.*, 2006). However, this methodology is a laborious and time-consuming process that requires large amounts of sample and this represents a major problem when it has a very limited amount of sample, which is the classic problem that occurs when trying to establish the profile of volatile organic compounds present in specimens obtained by micropropagation.

Moreover, when investigators extract essential oils from a plant matrix to characterize the profile of VOCs, little attention is paid to the possibility that the extraction methods may yield different essential oil profiles, or even worse, sample degradation, despite it being well known that chemical reactions can occur during the distillation process (Babu & Kaul, 2007).

Furthermore, it is a simple and fast modern tool which is used to characterize the volatile fraction of aromatic and medicinal plants (Vázquez *et al.* 2014), offering a valid alternative to the classic Clevenger essential oil hydrodistillation for gas chromatographic analysis of volatile constituents.

To our knowledge, there are no enough reports in the literature about direct SPME analysis of volatile constituents on whole plants of *C. gilliesii* or their aerial parts. The aim of this research was to evaluate the HS-SPME/GC-MS method for the analysis of volatile compounds in this specie. Additionally, a comparison between the results obtained by HS-SPME and essential oil analyses on the characteristic GC-MS profiles was performed for both qualitative and semi-quantitative analyses of the VOCs from *C. gilliesii*.

MATERIAL AND METHODS

Plant Samples

Specimens of *C. gilliesii* in the process of flowering-fruiting were collected in the Sierras Grandes de Córdoba, Argentina (35°32'384 S; 64°33'252 O; altitude: 1663 meter above sea level). A whole plant was deposited in the Herbarium Marcelino Sayago (Register Number UCCOR 402), Faculty of Agricultural Sciences, at Catholic University of Córdoba. To perform the analysis of HS-SPME/GC-MS

within 24 hs. of collection, samples (100.0 ± 0.1 mg) of aerial parts of fresh plants were previously chopped with a clean cutter and placed in glass vials of 20 cm^3 , which were sealed with Viton septa and aluminium seals provided by Supelco (Sigma-Aldrich, Argentina) (Figure 1).

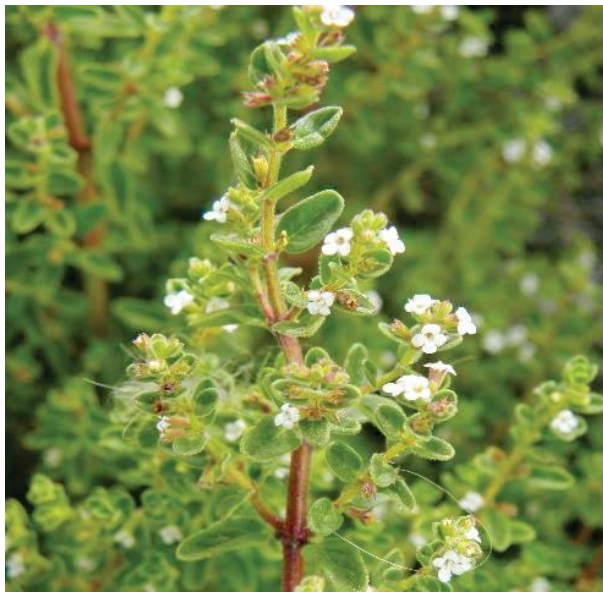


Figure 1. *Clinopodium gilliesii* (Benth.) Kuntze

HS-SPME technique

The analytical conditions for HS-SPME analysis were previously established by the measurement and characterization of the volatile compounds present in samples of *Clinopodium odorum* (Griseb), Harley (Vázquez *et al.*, 2014). Using a manual holder (Supelco), DVB-CAR-PDMS 50/30 μm fiber (Sigma-Aldrich, Argentina) was conditioned in the GC injector at 225°C for 8 hours before use.

The vials containing the samples of *C. gilliesii* were immersed in a thermostatic water bath at 40.0°C (PolyScience 8005, accuracy 0.2°C). After 10 min, the SPME device was inserted into the sealed vial by manually penetrating the septum, and the fiber was exposed to the sample headspace for 10 min. After extraction, the needle on the SPME manual holder was set to its maximum length in the GC injector and the fiber was directly exposed to the hot injector at 250°C for 5 min in splitless mode.

Essential oil of *C. gilliesii*

500 g of aerial parts of plants were hydrodistilled for three hours. The aqueous distillate was extracted with chloroform ($3 \times 20\text{ mL}$) and the organic layer was

separated, dried over anhydrous MgSO_4 and filtered. The solvent was evaporated in a rotary evaporator (ambient temperature) to obtain 2.3 mL of essential oil (yield of 0.46% v/w). An aliquot diluted in hexane of the essential oil in hexane was injected into the chromatograph.

Gas Chromatography-Mass Spectrometry

The identification of volatile components was performed using a gas chromatograph HP 5890 Series II equipped with a manual injection port operating in a split/splitless mode and coupled to an HP 5970 Mass Detector. The column used was an HP-5 capillary column ($30\text{ m} \times 0.25\text{ mm ID} \times 0.25\text{ }\mu\text{m}$ film). The working conditions were: injector: 225°C ; interface: 230°C , gas carrier: He 99.99%; head pressure: 5 psi; initial ramp: 40°C (5 min) - 90°C ($2^\circ\text{C}/\text{min}$); middle ramp: $90\text{-}130^\circ\text{C}$ ($10^\circ\text{C}/\text{min}$); final ramp: $130\text{-}200^\circ\text{C}$ ($5^\circ\text{C}/\text{min}$). The mass spectrometer was operated at 70 eV and the spectra were recorded in the range of m/z 30 - 550 amu in the acquisition mode "scan-full." The data processing system used was the HP-MS ChemStation including database Wiley 275 and NIST. The volatile components were identified by comparing their mass spectra with library data (match $\geq 90\%$) and by the determination of the respective Kovat's retention indices (KI), (alkane standards provided by Sigma-Aldrich).

Statistical analysis

All analysis was done in triplicates. Modified Shapiro-Wilk statistic was applied to test data normality. One-way analysis of variance (ANOVA, $\alpha = 0.05$) and Tukey's test were carried out on quantitative data, in order to verify if there were significant differences in the composition of the samples of the different aerial parts of the plants. Data were analyzed using INFOSTAT software Version 2016 (Facultad de Ciencias Agropecuarias, Universidad Nacional de Córdoba, Córdoba, Argentina).

RESULTS AND DISCUSSION

HS-SPME of leaves

As can be seen in Table 1, the existence of 69 different components in the volatile fraction of leaves of *C. gilliesii* were established, 67 of which were successfully identified (97.1%). Thus, positive identification was achieved in 98.8% of the total area observed in the chromatogram.

From measurements made on leaves of *C. gilliesii* (Table 1), the major components were piperitone oxide (16.1%) and piperitenone oxide (10.2%). Lower proportions were observed of bicyclogeremacrene (8.6%), δ -cadinene (4.4%), β -caryophyllen (4.3%), germacre-4-ol (4.3%), germacrene D (3.1%), bicycloelemene (3.0%),

linalool (2.6%), viridiflorol (2.5%), limonene (1.7%), aromadendrene (1.7%), β -bourbonene (1.6%), geranial (1.3%), dodecane (1.3%), γ -muurulene (1.2%), sabinyl acetate (1.1%), α -gurjunene (1.1%), α -copaene (1.1%), nerol (1.0%) and pulegone (1.0%). The rest of the components were present at amounts ranging from 0.9% (for example verbenone) to 0.2% (for example neryl acetate and spathulenol, among others). Additionally, a compound identified like (1-butenyl)thiophene by MS was observed in 3.0%.

HS-SPME of stems

As can be seen in Table 1, the existence of 26 different components in the volatile fraction of the stems of *C. gilliesii* was established and was successfully identified (100.0%). Thus, positive identification was achieved in 100.0% of the total area observed in the chromatogram.

In stems of *C. gilliesii* (Table 1), eucaliptol (43.4%) and β -pinene (8.9%) were observed to be the main components. All other compounds were found in appreciable amounts (>1%), for example: myrtenal (4.0%), limonene (3.6%), piperitone oxide (3.2%), menthone (3.0%), geranial (3.0%), myrtenol (2.3%), δ -cadinene (2.2%), germacrene-4-ol (2.1%), neral (1.9%), α -methylcinnamaldehyde (1.8%), linalool (1.5%), γ -cadinene (1.5%), *trans*-limonene oxide (1.4%), piperitenone oxide (1.3%), γ -terpinene (1.3%), epizonarene (1.2%), and bicyclogermacrene (1.1%), were observed. The rest of the components were present at amounts ranging from 0.8% (α -copaene) to 0.5% (*trans*-dihydrocarvone, among others). Additionally, a compound identified like (1-butenyl)thiophene by MS was observed in 5.7%.

HS-SPME of flowers

As can be seen in Table 1, the existence of 63 different components in the volatile fraction of the inflorescence of *C. gilliesii* were established, 62 of which were successfully identified (98.4%). Thus, positive identification was achieved in 99.6% of the total area observed in the chromatogram.

Piperitenone oxide (12.8%) was found to be the major components. Additionally, appreciable amounts of bicyclogermacrene (6.1%), limonene (5.9%), germacrene-4-ol (4.5%), δ -cadinene (4.3%), β -caryophyllene (3.7%), piperitone oxide (3.4%), bicycloelemene (2.9%), pulegone (2.7%), geranial (2.5%), δ -cadinene (2.2%), verbenone (2.0%), were found. The presence of β -bourbonene (1.7%), viridiflorol (1.6%), piperitenone (1.4%), *p*-cymenene (1.4%), γ -muurulene (1.3%), thymol (1.2%), isoeledene (1.1%), *trans*- β -farnesene (1.1%), and diosphenol (1.1%), also were observed. The rest of the observed components were present at amounts

ranging from 0.9% (α -terpineol) to 0.2% (agarospirol). Additionally, a compound identified like (1-butenyl)thiophene by MS was observed in 4.3%.

HS-SPME of combined aerial parts of the whole plant

As can be seen in Table 1, the existence of 59 different components in the volatile fraction of *C. gilliesii* were established, 58 of which were successfully identified (98.3%). Thus, positive identification was achieved in 99.9% of the total area observed in the chromatogram.

Table 1 summarizes the main components provided by the HS-SPME analysis of the aerial parts of the whole plant, with piperitenone oxide (20.0 %), germacrene-4-ol (11.5%), bicyclogermacrene (9.2%) and piperitone oxide (7.3%) being found at the greatest proportions. In addition, there were also appreciable amounts of δ -cadinene (5.2%), germacrene D (4.5%), viridiflorol (3.3%), β -caryophyllene (3.2%), bicycloelemene (2.4%), γ -muurulene (1.3%), geranial (1.2%), and *trans*- β -farnesene (1.1%). The rest of the observed components were present at amounts ranging from 1.0% (aromadendrene) to 0.1% (thymol). Additionally, a compound identified like (1-butenyl)thiophene by MS was observed in 3.0%.

Essential oil analysis

As can be seen in Table 1, the existence of 88 different components in the essential oil obtained by hydrodistillation of desiccated aerial parts of *C. gilliesii* were established, 81 of which were successfully identified (92.1%). Thus, positive identification was achieved in 97.1% of the total area observed in the chromatogram.

The major components present in the essential oil of *C. gilliesii* were piperitone oxide (18.1%), and piperitenone oxide (17.4%). There were also significant amounts of: bicyclogermacrene (4.6%), δ -cadinene (4.4%), α -muurulol (4.3%), δ -cadinol (3.5%), germacrene-4-ol (3.5%), spathulenol (2.9%), linalool (2.7%), geranial (2.6%), germacrene D (2.3%), neral (1.7%), β -caryophyllene (1.8%), pulegone (1.3%), α -copaene (1.1%), α -fenchyl acetate (1.1%), and nerolidol (1.0%). Other components were found at amounts ranging from 0.9% (myrtenal) to 0.1% (α -cubebene among others). It is remarkable that the compound (1'-butenyl)thiophene was present in 0.3%.

Differences between the composition of flowers, stems and leaves by SPME analysis

One of the most advantages of SPME is the low amount of sample required for the realization of analytical determinations. On the other hand, because of the low yield obtained, higher amounts of sample are necessary for the

production of essential oils by plants hydro distillation. Typically, 0.5 - 1 kg of sample is required to obtain a suitable sample for performing the analyses (Viturro *et al.*, 2007). This difference makes it possible, in SPME analyses, to perform quickly and easily the analytical data from the different parts of the plant, whereas is difficult the separation in different parts (stems, leaves and flowers) for later essential oil obtaining. However, many scientific works have been published lately, where the different parts of plants were processed separately, founding concur with our observations, significant differences in chemical composition of them (Kothari *et al.*; 2005; Angioni *et al.*; 2006; Goel *et al.*; 2007; Radulović *et al.*; 2009; Mohammadhoseini *et al.*; 2013; Santos *et al.*; 2013; Mohammadhoseini, 2015 a and b; Rajeswara Rao *et al.*; 2015; Kovacevic *et al.*; 2016; Mohammadhoseini *et al.*; 2016; Shahid Ud-Daula *et al.*; 2016, y Wesolowska & Jadczyk, 2016).

Whole plant essential oil obtained by hydro distillation studies have been reported in the particular case of *C. gilliesii* (Viturro *et al.*, 2007). Currently, no study on the contribution of each aerial part of the plant to the set of VOCs produced by *C. gilliesii* has been informed. In consequence, this report could represent the first description of the VOCs observed in each aerial part of the plant separately. The data summarized in Table 1 show some interesting differences between the results of the HS-SPME analysis of flowers, stems and leaves of *C. gilliesii*:

*There was a significant difference in the number of compounds produced by each part of the plant. Both the flowers and the leaves were responsible for the greatest number of volatile organic compounds (69 and 63 respectively), while the stems contribute a smaller number (only 26 compounds).

*The main components piperitone oxide and piperitenone oxide were found at greater proportions in leaves and inflorescence than in stems. However, the flowers were the main source of piperitenone oxide and the leaves were the main source of piperitone oxide.

*The contribution of β -pinene, menthone, myrtenal, myrtenol, nerol, neral, geranial, and epizonarene, were primarily due to the stems while bicycloelemene, β -caryophyllene, germacrene D, bicyclgermacrene, δ -cadinene and viridiflorol were supplied solely by the leaves. The contribution of limonene, linolool, verbenone, pulegone, piperitenone and γ -cadinene were supplied solely by the flowers.

*Moreover, eucalyptol and trans-dihydrocarvone were observed exclusively in steams and β -phellandrene, 4-terpineol, trans-carveol, anisole, geranyl acetate, α -elemene, and spathulenol were exclusive component of the leaves.

HS-SPME analysis of whole plant vs. essential oil analysis

Using HS-SPME analysis in the previously established conditions, the existence of 59 different components in the volatile fraction of *C. gilliesii* were established, while in the essential oil analysis 89 components were determined (see Table 1). This situation represents an observation of 34% over components using essential oil analysis.

Moreover, the data summarized in Table 1, show some differences between the results of the HS-SPME analysis of the whole plant and those from the essential oil. Comparing the essential oil and HS-SPME data revealed main components not be the same. In HS-SPME analysis, piperitenone oxide (20.0%) was predominant while piperitone oxide (18.1%) and piperitenone oxide (17.4%) were present in approximately the same amount in the essential oil. Moreover, in HS-SPME analysis of whole plant, appreciable quantities of germacrene-4-ol (11.5%), bicyclgermacrene (9.2%), and germacrene D (4.5%) were observed while these compounds were present in a minor proportion in the essential oil (3.2%, 4.6%, and 2.3% respectively). Additionally, appreciable amounts (>1%) of α -fenchyl acetate and nerolidol were observed only in essential oil analysis.

It is interesting to note also that the greatest differences observed between the results obtained by SPME and essential oil can be seen at long retention time in the chromatogram (time > 39 min). In this case, the major differences were in minority component present only in the essential oil and not in the SPME analysis. In this sense, α -elemene (0.3%), γ -amorphene (0.3%), α -farenesene (0.5%), cadina-1,4-diene (0.2%), nerolidol (1.0%), 1-endo-bourbonanol (0.5%), β -oplopenone (0.2%), τ -cadinol (0.1%), τ -muurolol (0.1%), β -eudesmol (0,2%), and cis-trans-farnesol (0.8%) were only present in the essential oil analysis.

This observed difference between the results obtained by HS-SPME analysis of fresh aerial parts and the essential oil analyses just could perhaps be explained through enzymatic processes and microbiological changes occurred during the drying process (Vázquez *et al.*, 2014). On the other hand, it is important to take into account that, except for the components found only by HS-SPME and not in the essential oil, the differences could be due to differences in the affinities of the components by the fiber, thi is, to differences in absorption / desorption processes. However, more studies are necessary by support this hypothesis.

Additionally it should be noted that the results presented here, the number of components described was much higher than those reported previously by Viturro *et al.* (2007).

Table 1. Volatile compounds observed in *Clinopodium gilliesii* (Benth.) Kuntze

Pea k	Rt ^a (min)	Compound ^b	KI ^c (Ex)	KI ^d (Lit)	% ^e				
					L ^f	S ^g	F ^h	WP ⁱ	EO ^j
1	16.500	β-Pinene	965	969	0.3a	8.9b	0.4a	0.4	0.2
2	18.010	β-Myrcene	987	983	0.4a	ND	0.7b	0.4	0.1
3	19.372	Sabinene	1006	1001	0.6a	ND	ND	0.2	ND
4	20.001	2,3,4,5-Tetrahydroanisole	1015	1015	ND	ND	0.5a	ND	ND
5	20.464	Limonene	1021	1025	1.7a	3.6b	5.9c	0.4	0.3
6	20.577	Eucalyptol	1023	1023	ND	43.4a	ND	0.5	0.4
7	21.426	β-Phellandrene	1034	1035	0.3a	ND	ND	ND	0.1
8	22.204	<i>trans</i> -β-Ocimene	1044	1044	0.8a	ND	0.7a	0.3	0.2
9	22.827	γ-Terpinene	1053	1050	ND	1.3a	ND	ND	ND
10	25.010	p-Cymenene	1082	1085	ND	ND	1.4a	ND	ND
11	26.104	Linalool	1096	1093*	2.6b	1.5a	2.7b	1.4	2.7
12	28.469	Myrcenol	1133	1126	ND	ND	ND	ND	0.3
13	29.599	<i>trans</i> -Limonene oxyde	1143	1143	ND	1.4a	ND	ND	ND
14	29.695	Isomenthone	1153	1153	ND	ND	ND	ND	0.1
15	30.693	Menthone	1168	1166	0.4a	3.0b	0.2a	ND	0.2
16	30.730	Lavandulol	1169	1170	ND	ND	ND	ND	0.1
17	31.301	<i>cis</i> -Isopulegone	1177	1173	0.5a	ND	0.3a	0.3	0.3
18	31.516	4-Terpineol	1181	1182	0.7a	ND	ND	ND	0.4
19	31.709	Isopinocarveol	1185	1183	ND	ND	ND	ND	0.1
20	32.041	<i>trans</i> -Isopulegone	1189	1188	0.5a	ND	0.6a	0.5	0.9
21	32.207	Myrtenal	1192	1193	0.7a	4.0b	0.6a	0.2	0.9
22	32.394	Myrtenol	1195	1196	0.2a	2.3b	ND	ND	0.6
23	32.523	α-Terpineol	1197	1196	0.3a	ND	0.9b	0.8	ND
24	32.638	Dodecane	1199	1200	1.3a	ND	ND	ND	ND
25	32.774	<i>trans</i> -Dihydrocarvone	1203	1201	ND	0.5a	ND	0.2	0.6
26	33.066	<i>trans</i> -Carveol	1211	1210	0.5a	ND	ND	0.2	ND
27	33.129	Verbenone	1214	1212	0.9a	ND	2.0b	0.8	ND
28	33.246	4,7-Dimethylbenzofuran	1218	1218	ND	ND	ND	ND	0.8
29	33.480	α-fenchyl acetate	1225	1226	ND	ND	ND	ND	1.1
30	33.597	Nerol	1229	1229	1.0b	1.5c	0.7a	0.8	0.3
31	33.894	Pulegone	1236	1233*	1.0b	0.6a	2.7c	0.9	1.3
32	34.001	Neral	1242	1242	0.9a	1.9b	0.9a	0.7	1.7
33	34.389	Piperitone Oxide	1253	1259	16.1b	3.2a	3.4a	7.3	18.1
34	34.672	Carvone oxide	1263	1263	0.4a	ND	0.6a	0.8	0.4
35	34.750	Anisole	1265	1265	0.2a	ND	ND	0.5	ND
36	34.815	Diosphenol	1267	1273	0.4a	ND	1.1b	ND	0.6
37	34.936	Geranial	1271	1271	1.3a	3.0c	2.5b	1.2	2.6
38	35.190	Citronellyl formiate	1279	1282	ND	ND	ND	ND	0.3
39	35.209	Bornyl acetate	1280	1280	0.3a	ND	0.7b	0.1	0.1
40	35.441	(1'-butenyl)thiophene	1287		3.0a	5.7c	4.3b	3.0	0.3
41	35.635	Sabinyl acetate	1290	1291	1.1a	ND	0.9a	ND	0.3
42	35.647	Unknown	1294		ND	ND	ND	ND	0.1
43	35.718	Unknown	1296		ND	ND	ND	ND	0.5
44	35.762	Thymol	1297	1295*	0.9b	ND	1.2c	0.1	0.8
45	36.033	Unknown	1307		0.4a	ND	0.4a	ND	ND

Tabla 1. Continuation

Pea k	Rt ^a (min)	Compound ^b	KI ^c (Ex)	KI ^d (Lit)	L ^f	S ^e	% ^e F ^b	WP ^d	EO ^j
46	36.167	Unknown	1312		0.8a	ND	ND	0.1	0.7
47	36.440	Myrtenyl acetate	1322	1322	ND	ND	ND	ND	0.9
48	36.492	δ-Elemene	1324	1325	0.6b	ND	0.9c	0.4	ND
49	36.600	α-Methylcinnamaldehyde	1328	1330	0.2a	1.8b	0.4a	0.3	ND
50	36.805	Bicycloelemene	1335	1334	3.0a	ND	2.9a	2.4	0.6
51	36.962	Piperitenone	1342	1339	0.5a	ND	1.4b	0.8	0.3
52	37.138	α-Cubebene	1346	1345	0.3a	ND	0.9b	0.1	0.1
53	37.239	Citronellyl acetate	1352	1353	0.5a	ND	0.3a	0.1	0.8
54	37.523	Neryl acetate	1363	1364	0.2a	ND	0.5b	ND	0.3
55	37.680	Piperitenone Oxide	1369	1369	10.2b	1.3a	12.8c	20.0	17.4
56	37.742	Isoledene	1371	1373	ND	ND	1.1a	ND	ND
57	37.880	α-Ylangene	1377	1377	0.6a	ND	0.9b	0.7	ND
58	38.016	Geranyl acetate	1382	1383	0.2a	ND	ND	0.2	0.6
59	38.122	β-Bourbonene	1386	1385	1.6a	ND	1.7a	0.8	0.5
60	38.288	β-Elemene	1392	1391	0.8a	ND	0.9a	0.8	0.4
61	38.401	α-Copaene	1396	1400	1.1b	0.8a	0.7a	0.8	1.1
62	38.549	cis-Jasmone	1402	1403	0.5a	ND	1.6b	0.4	0.7
63	38.707	Isolongipholene	1409	1406	ND	ND	ND	ND	0.1
64	38.774	β-Maaliene	1411	1411	0.3a	ND	0.3a	0.1	0.1
65	39.039	β-Caryophyllene	1421	1418	4.3c	0.6a	3.7b	3.2	1.6
66	39.268	β-Cubebene	1432	1432	0.9a	ND	0.9a	0.7	0.1
67	39.427	γ-Elemene	1437	1433	0.5a	ND	0.6a	0.2	0.2
68	39.539	Aromadendrene	1443	1446	1.7a	ND	1.7a	1.0	0.2
69	39.645	α-Gurjunene	1448	1449	1.1a	ND	1.2a	0.9	0.2
70	39.804	trans-β-Farnesene	1454	1451	0.4a	ND	1.1b	1.1	0.9
71	39.902	α-Elemene	1458	1460	0.7a	ND	ND	ND	0.3
72	40.128	γ-Muurolene	1468	1469	1.2a	ND	1.3a	1.3	0.1
73	40.378	epi-Bicyclosesquiphellandrene	1476	1471	0.2a	ND	0.2a	ND	0.1
74	40.455	α-Amorphene	1482	1481	0.8a	ND	0.8a	0.7	0.2
75	40.593	Germacrene D	1487	1485	3.1b	ND	2.3a	4.5	2.3
76	40.630	γ-Amorphene	1489	1488	ND	ND	ND	ND	0.3
77	40.798	Epizonarene	1496	1497	0.4a	1.2b	1.0b	0.7	0.1
78	40.998	Bicyclogermacrene	1503	1499	8.6c	1.1a	6.1b	9.2	4.6
79	41.064	α-Farnesene	1507	1509	ND	ND	ND	ND	0.5
80	41.214	β-Bisabolene	1513	1514	0.2a,b	ND	0.3b	0.8	0.1
81	41.391	γ-Cadinene	1521	1524	2.4b	1.5a	2.8c	3.2	0.5
82	41.576	δ-Cadinene	1527	1525	4.4b	2.2a	4.3b	5.2	4.6
83	41.633	Cadina-1,4-diene	1531	1532	ND	ND	ND	ND	0.2
84	41.816	α-Cadinene	1538	1538	0.2a	ND	0.2a	ND	0.1
85	41.938	Elemol	1543	1542	0.5a	ND	0.6a	0.8	0.2
86	42.152	Unknown	1552		ND	ND	ND	ND	0.2
87	42.420	Nerolidol	1563	1562	ND	ND	ND	ND	1.0
88	42.602	Germacren-4-ol	1571	1576	4.3b	2.1a	4.5b	11.5	3.5
89	42.860	1-endo-Bourbonanol	1581	1574	ND	ND	ND	ND	0.5
90	42.944	Spathulenol	1582	1582	0.2a	ND	ND	0.7	2.9
91	43.080	Viridiflorol	1590	1590	2.5c	1.6a	2.0b	3.3	0.8

Tabla 1. Continuación

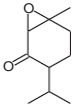
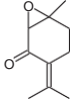
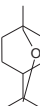
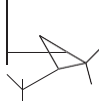
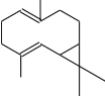
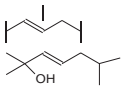
Pea k	Rt ^a (min)	Compound ^b	KI ^c (Ex)	KI ^d (Lit)	% ^e				
					L ^f	S ^g	F ^h	WP ⁱ	EO ^j
92	43.335	Unknown	1601		ND	ND	ND	ND	1,1
93	43.517	β -Oplophenone	1609	1608	ND	ND	ND	ND	0.2
94	43.639	Unknown	1614		ND	ND	ND	ND	0.2
95	43.796	Agarospirol	1621	1620	ND	ND	0.2a	ND	0.1
96	43.919	τ -Cadinol	1627	1625	ND	ND	ND	ND	0.1
97	44.050	τ -Muurolol	1633	1632	ND	ND	ND	ND	0.2
98	44.136	Unknown	1636		ND	ND	ND	ND	0.1
99	44.387	δ -Cadinol	1647	1645	0.2a	ND	0.2a	0.5	3.5
100	44.494	α -Cadinol	1652	1556	0.1a	ND	0.2a	0.5	1.6
101	44.609	β -Eudesmol	1657	1658	ND	ND	ND	ND	0.2
102	44.747	α -Muurolol	1663	1666	ND	ND	0.2a	ND	4.3
103	45.553	<i>cis-trans</i> -Farnesol	1699	1697	ND	ND	ND	ND	0.8
Total					100.0	100.0	100.0	100.0	100.0
% Compound Identified					97.1	100.0	98.4	98.3	92.1
% Area Identified					98.9	100.0	99.6	99.9	97.1

^aRt: retention time; ^b Identified by GC-MS; ^c Experimental Kovat's Retention Index; ^d Kovat's Retention Index from the Literature; ^e Means followed by different letters in each row denote significant differences between samples of different aerial parts of plants (ANOVA and Tukey's test, $\alpha = 0.05$); ^f Leaves; ^g Stems; ^h Flowers; ⁱ Whole Plant; ^j Essential Oil.

*RI Determinate using a standard.

ND = Not detected under the adopted conditions.

Table 2. Main volatile components found in samples

Compound	Chemical structure	Main uses and biological activities ^a
Piperitone oxide		Antibacterial and antifungal activities.
Piperitenone oxide		Larvicidal, ovicidal, oviposition-deterrent, developmental toxicity, and repellent properties. Flavouring agent.
Eucaliptol		Mucolytic, bronchodilating and anti-inflammatory properties. Antibacterial and antifungal activities. Flavouring agent.
β -Pinene		Antioxidant and antimicrobial activities. Flavouring agent in cleaning and furnishing care products.
Bicyclogermacrene		Insecticidal and larvicidal properties. Antioxidant, antimicrobial and antifungal activities.
Germacrene-4-ol		Antibacterial and insecticidal activities.

CONCLUSION

Using very smaller samples, a shorter extraction time and a much simpler procedure, the HS-SPME method can achieve comparable results to those obtained by essential oil analysis. Additionally, HS-SPME method used here could be employed to make quality control of commercial samples allowing the composition of volatile organic compounds from the separate aerial parts of *C. gilliesii* to be rapidly determined.

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